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# Fabrication of biocomposites reinforced with natural fibers and evaluation of their physiochemical properties

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# ABSTRACT

The present study aims to fabricate and evaluate the physio-chemical features of biocomposites filled with natural resources. Kenaf was used as a natural resource model and prepared by mechanical milling and sieving to obtain short fiber form. Virgin and recycled acrylonitrile butadiene styrene (ABS) were used as biocomposite matrix. Granular biocomposite was fabricated using the single screw extruder with the natural filler content variations ranging from 10 to 15%. Chemical composition, density, and dimension of the fabricated fillers were examined. Melt flow index, surface free energy, and functional group characteristics of the presently fabricated biocomposites were also evaluated. Analysis of the chemical composition, density, and dimension confirmed the physio-chemical characteristics of the fabricated fillers. The biocomposites density and melt flow index were slightly enhanced with the rise of the filler content. Contact angle, dispersive component, polar component, and surface free energy of biocomposites varied depending on the filler content and ABS type. Fourier transforms infrared spectroscopy analysis confirmed the functional groups of the biocomposites. In general, this work has successfully fabricated and evaluated the physio-chemical characteristics of composites filled with kenaf. The produced biocomposites can be a good strategy to produce high performance polymer biocomposites for future applications.

Keywords: Biocomposites; acrylonitrile butadiene styrene; surface free energy.

#### 1. INTRODUCTION

Development of biomaterials is rapidly expanding and has the potential prospect for the future [1-6]. This is because natural resources as basic reinforcements are abundantly available [7]. In addition, there are evidence in scientific literature that the performance of composites filled with natural resources is comparable with those filled with synthetic materials [8,9]. In addition, the use of natural filler has several advantages such as low density, low cost, renewable, and recyclable [10]. *Hibiscus cannabinus* is a natural resource that is widely explored as a filler for fabricating biocomposites because of its high cellulose content.

A combination of natural fibers and polymers provides a positive impact on the environments such as reducing the release of carbon gas and environmental impact of non-biodegradable material. Acrylonitrile butadiene styrene (ABS) is a type of polymer that is commonly explored as a matrix to produce biocomposites [11]. ABS becomes popular since it has been proven to have superior mechanical properties, chemical resistance, easily processed, and recyclable [12]. In general, a coupling agent is required to bind between nonpolar and polar materials such as polymers and cellulose, respectively.

Several studies have explored the novel properties of biocomposites for diverse field applications [13-16]. For instance, the chitosan-smectite biocomposite fabricated using the intercalation of protonated chitosan molecules into the interlayer space of smectite through cation exchange mechanism was proposed for the novel potentiometric monohydrogen phosphateselective sensor [17]. Their study verified that the biocomposites exhibited good performances in terms of sensitivity, stability, response time, detection limit, and a wide linear range for monohydrogen phosphate ions detection. Alternatively, the chitosan-lysozyme biocomposites exhibited the capability as an adsorbent to remove methyl orange dye and hexavalent chromium ions by maximum adsorption capacities of 435 and 216 mg/g, respectively [18]. Moreover, the biocomposites reinforced by fibers have also been widely explored for medical applications such as tissue engineering or regenerative medicine [19].

The properties of biocomposites can be influenced by several factors such as filler characteristics, filler loading, preparation procedure, and interface properties of filler in their matrix. Fabrication of biocomposites commonly involves the role of heat and mechanical such as milling, extrusion, and injection molding. Therefore, melt flow index determination is an important aspect to characterize the ability of biocomposites to load heat. Treatment on the surface of composites such as coating can affect the adhesion properties of biocomposites. Surface free energy is a criterion that is widely used to determine adhesion properties of biocomposites. In addition, the properties of biocomposites are also highly influenced by their internal structure. Each molecule in biocomposites has a natural vibration, which depends on the type and chemical bonding. Natural vibration of molecules is an indicator for the identification of molecules present in the materials. Identification of functional group of biocomposites can be easily performed using the infrared spectrophotometers such as Fourier transform infrared spectroscopy. From the above mentioned overview, it is possible to fabricate a few trends for the behavior of polymer matrix composites based on the nature of the polymer matrix and filler.

Aligning the aforementioned research necessity, this work was aimed to fabricate and evaluate the physico-chemical features

#### 2. MATERIALS AND METHODS

## 2.1. Materials.

Kenaf having a length of 4 m was obtained from the kenaf plantation under PT Global Agrotek Nusantara, Pekanbaru, Riau, Indonesia. Virgin and recycled ABS polymers were purchased from PT MUB Jaya, Bogor, Indonesia. Additive chemicals were maleic acid (Darmstadt, Germany) used as a coupling agent and primary antioxidant (Zaozhuang, China).

#### 2.2. Short fibers fabrication.

Kenaf fibers were processed by soaking dynamically in the water for two weeks and then dried under the sun. Furthermore, the fibers were cut with a uniform length of  $\pm 1$  cm and then dried in an oven (Tipe YNC-OV, YENACO, China) at 40 °C for 24 h. The dried kenaf was taken from the oven and then milled using a milling machine (Model MDY-1000, FOMAC, China). After the milling process, the sample was screened by using 20 mesh sieve and was selected as short fibers. The chemical composition of fiber was determined according to the technical association of the pulp and paper industry (TAPPI) standardization. Kenaf fiber and biocomposites densities were estimated using the Archimedes approach. Dimension of the fiber was estimated from images (Model BX51, Olympus, Japan) combined with the software olympus DP2-BSW and DP25 olympus microscope camera.

#### 2.3. Preparation of biocomposites.

Granular biocomposite was fabricated by mixing kenaf short fiber as a filler, ABS polymer as a matrix, maleic acid as coupling agent, and primary antioxidant with the biocomposite composition as listed in Table 1.

Sample	Kenaf short fiber	ABS polymer	Primary antioxidant	Maleic acid
vABS-KSF-	10 %	87 %	1%	2%
10%				
vABS-KSF-	15 %	82 %	1%	2%
15%				
rABS-KSF-	10 %	87 %	1%	2%
10%				
rABS-KSF-	15 %	82 %	1%	2%
15%				

**Table 1.** Composition of the presently fabricated biocomposites

All samples were fabricated using the single screw extrusion (Model HXSJ-125/125, Kai Xin, China). Test pieces were prepared using the molding injection machine (Model HC-250, Hwa Chin, China) In the preparation, 5 kg of granular biocomposite was put into the hopper of molding injection and then heated at a

#### **3. RESULTS**

#### 3.1. Physio-chemical characteristics of fiber.

All natural fibers primarily contain cellulose, hemicellulose, lignin and extractives. Table 2 shows the chemical composition of the presently fabricated fibers. This study found that the cellulose, lignin, holoscellulose, hemicellulose, and extractives contents present in the fibers were 66.47, 2.39, 75.43, 9.43, and 2.11%,

of biocomposites filled with natural resources. Specifically, ABS and kenaf were employed as a matrix and a filler for the presently fabricated biocomposites, respectively. Outcomes from this work are highly beneficial in improving the biocomposites properties that are important for future applications.

temperature of 80 °C. Barrel of the molding injection was heated at 5 different temperatures (120 °C to 200 °C). Therefore, vABS-KSF-10%, vABS-KSF-15%, rABS-KSF-10%, and rABS-KSF-15% denote to virgin ABS with 10% of filler, virgin ABS with 15% of filler, recycled ABS with 10% of filler, and recycled ABS with 15% of filler, respectively.

#### 2.4. Melt flow index investigation.

Melt flow index (MFI) is a measure of polymer melting rate expressed by the weight of polymer in gram run for 10 mins through the capillaries. Biocomposites MFI measurements were performed using the melt flow indexer (Model XNR-400D, Jinan Hensgrand Instrument, Cina) at a temperature of 220 °C.

#### 2.5. Fourier transforms infrared spectroscopy analysis.

Fourier transforms infrared spectroscopy (FTIR) analysis was conducted by measuring the transmittance of materials using the FTIR equipment (ABB, model MB300, Canada) observed within the wavenumber range from 4000 cm<sup>-1</sup> to 400 cm<sup>-1</sup>. The sample was formed into pellets with potassium bromide.

# 2.6. Surface free energy determination.

Contact angle measurement was observed through the sessile droplet method using Phoenix 300 contact angle analyzer (Surface Electro Optics, Korea) and was analysed based on equation [20]:

$$\gamma_l \cos \theta = \gamma_s - \gamma_{sl} \tag{1}$$

where  $\gamma_l$  denotes the surface free energy of liquid in contact with the solid and  $\gamma_{sl}$  represents the interfacial free energy between solid and the liquid. In order to solve this equation for  $\gamma_s$ , it needs to add another equation that correlates  $\gamma_{sl}$  with  $\gamma_l$  and  $\gamma_s$ . It was proposed that the polar interaction could be computed using the same geometric mean mixing rule as for the dispersion force interaction [21]. If the contact angle of at least two liquids (polar and nonpolar), with known  $\gamma_l^d$  and  $\gamma_l^p$  parameters are measured on a solid surface,  $\gamma_s^d$  and  $\gamma_s^p$  parameters of that solid can be estimated. Hence, the following equation was used:[21]

$$\boldsymbol{\gamma}_{sl} = \boldsymbol{\gamma}_s + \boldsymbol{\gamma}_l - 2\left(\boldsymbol{\gamma}_s^d \boldsymbol{\gamma}_l^d\right)^{1/2} - 2\left(\boldsymbol{\gamma}_s^p \boldsymbol{\gamma}_l^p\right)^{1/2} \tag{2}$$

By substituting equation (2) into Young's equation (1), the equation (1) becomes:

$$(1+\cos\theta)\gamma_l = 2\left(\gamma_s^d\gamma_l^d\right)^{1/2} + 2\left(\gamma_s^p\gamma_l^p\right)^{1/2}$$
(3)

respectively. In general, it was obvious that the cellulose content was higher compared to other compositions. It was well known that the chemical composition of natural fibers highly depends on several parameters such as the plant origin and growth conditions as reported in scientific literature. For instance, an analysis of the chemical composition of the raw flax fibers exhibited that the

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moisture content, extractives, lignin, hemicellulose, and cellulose were 9.2, 6.2, 3.9, 15.7, and 64.8%, respectively [22].

Table 2. Chemical composition of the kenaf fiber					
Chemical composition	Percentage (%)				
Cellulose	66.47				
Lignin	2.39				
Holoscellulose	75.43				
Hemicellulose	9.43				
Extractives	2.11				

In addition, the density of the fiber is presented in Table 3. It was estimated that the fiber density was 1.00 g/cm<sup>3</sup>. It was reported from the previous study that the density of natural fibers ranged from 0.9 to 1.5 g/cm<sup>3</sup>. For instance, the extracted Alfa fibers have a density of 1.40 g/cm<sup>3</sup> [23]. In addition, the oil palm empty fruit bunch fibre has a density of 0.94 g/cm<sup>3</sup>. In general, it has been reviewed that the density of various natural fibers such as flax, hemp, jute, and harakeke are 1.5, 1.5, 1.3-1.5, and 1.3, respectively [24]. Moreover, it was also found that the length and diameter of the presently produced fibers were 897.07  $\mu$ m and 66.38  $\mu$ m, respectively, falling under microfiber category.

## 3.2. Biocomposites density.

Table 3 presents density of the presently produced biocomposites. Density of the biocomposite produced by mixing virgin ABS with 10% of filler was 1.067 g/cm<sup>3</sup>. A slight increase in the density for the biocomposite produced by mixing virgin ABS with 15%, which is 1.098 g/cm<sup>3</sup>. Alternatively, the biocomposite density fabricated by mixing recycled ABS with 10% of filler was 0.948 g/cm<sup>3</sup>. Also, a slight increase in the density for the biocomposite density fabricated by mixing recycled ABS with 15% of filler was 0.986 g/cm<sup>3</sup>.

Sample	Density (g/cm <sup>3</sup> )
Kenaf fiber	1.00
vABS-KSF-10%	1.06
vABS-KSF-15%	1.09
rABS-KSF-10%	0.94
rABS-KSF-15%	0.98

Table 3. Density of the kenaf fiber and biocomposites

#### 3.3. Melt flow index of biocomposites.

Table 4 presents the melt flow index of the produced biocomposites. It was found that the melt flow index of the biocomposite produced by mixing with virgin ABS with 10% of filler was 6.5 g/10 min. An increase in the melt flow index of the biocomposite produced by mixing with virgin ABS with 15% was observed by 9.5 g/10 min.

Table 4. Melt flow index of the biocomposites

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Sample	Melt Flow Index (g/10			
	min)			
vABS-KSF-10%	6.50			
vABS-KSF-15%	9.50			
rABS-KSF-10%	11.00			
rABS-KSF-15%	19.50			

A similar finding was also obtained for the biocomposite produced by mixing with recycled ABS. Enhancement of the value from 11.0 to 19.5 g/10 min can be obtained by increasing the filler contents from 10 to 15%. In general, the melt flow index of all produced biocomposites increased with increasing filler contents. The melt flow index for recycled ABS was generally higher compared to virgin ABS.

An increase in the melt flow index of the biocomposites would help their injection moldability. It is possible to allow the addition of greater amounts of the natural fillers when compared to virgin ABS. Moreover, it was reported that the melt flow index characteristics of natural fibers are also influenced by their cellulose contents [25].

# 3.4. Surface free energy analysis.

Surface free energy is one of thermodynamic quantities that describes equilibrium state of atoms in surface layer of material. It is a combination of polar and nonpolar (dispersive) energy [26]. Surface free energy of the presently fabricated was estimated using the contact angle value. Surface free energy estimation of the biocomposites is presented in Table 5.

 
 Table 5. Contact angle, dispersive component, polar component, and surface free energy of the biocomposites

Sample	Contact angle		Dispersive	Polar	Surface
	Water	Hexane	component	component	free energy
vABS-	80.71	14.04	17.85	10.73	28.58
KSF-					
10%					
vABS-	94.85	16.28	17.67	4.03	21.70
KSF-					
15%					
rABS-	69.47	15.10	17.77	18.20	35.97
KSF-					
10%					
rABS-	68.92	17.55	17.55	18.75	36.30
KSF-					
15%					

Table 5 shows that the contact angle values of the biocomposites fabricated from the virgin ABS tested by the water increase with the rise of the filler contents of 10 and 15% by 80.71 and 94.85°, respectively. A similar finding was also observed for the biocomposites when tested using hexane by increasing from 14.04 and 16.28° for the filler contents of 10 and 15%, respectively. A slight difference of the contact angle characteristics was observed when the biocomposites were produced from the recycled ABS. It was found that the contact angle tested using the water was insignificantly decreased from 69.47 and 68.92° for the filler contents of 10 and 15%, respectively. Increasing the contact angle value indicates an increase in the hydrophobic nature of the biocomposites. It can also be suggested that enhancement of the contact angle values are related to the increase in surface roughness of the biocomposites.

Figures 1 and 2 illustrated the liquid droplet test for the produced biocomposites using the water and hexane for polar and dispersive components investigations, respectively. It was found that the polar component of the biocomposites produced from virgin ABS was decreased by 10.73 and 4.03 for the filler contents of 10 and 15%, respectively. A decrease in the dispersive component was also observed for the biocomposites by 17.85 and 17.67 for the corresponding filler contents. In addition, the polar component of the biocomposites produced ABS were 18.20 and 18.75 for the filler contents of 10 and 15%, respectively. Moreover, the dispersive component of the biocomposites for the corresponding filler contents of 10 and 15%, respectively. Moreover, the dispersive component of the biocomposites for the biocomposites for the corresponding filler contents was 17.77 and 17.55.



Figure 2. Dispersive component test for the produced biocomposites

In general, the surface free energy of the biocomposites produced from virgin ABS was decreased by 28.58 and 21.70 for the filler contents of 10 and 15%, respectively. Alternatively, the surface free energy of the biocomposites produced from recycled ABS was slightly increased by 35.97 and 36.30 for the corresponding filler contents. The polar component of the biocomposites was responsible for increasing their surface energy.



Figure 3. FTIR spectra of (a) kenaf fibers, (b) virgin ABS, (c) vABS-KSF-10%, and (d) vABS-KSF-15%.

## 3.5. FTIR inspection.

FTIR analysis was conducted to determine the changes in functional groups of the biocomposites. Atomic and molecular compositions of the fibers and matrix surface are different. When the surface of fibers and the matrix bonds, it forms a specific chemical bond. The atoms in molecule vibrate to describe the vibrational energy levels. If the molecule absorbs infrared radiation, it would be excited to its higher level. Cellulose molecules have a strong tendency to form hydrogen bonds intra- and intermolecularly.

FTIR spectra are shown in Figures 3 and 4 show several peaks, which are at 3410 cm<sup>-1</sup>, 2901 cm<sup>-1</sup>, 1443 cm<sup>-1</sup>, and 1057 cm<sup>-1</sup> due to hydrogen bonding (OH), C-H, CH<sub>2</sub>, and C-O-C or C-O stretching. In addition, Figures 3 and 4 present the infrared absorption for the virgin and recycled ABS, respectively. Acrylonitrile monomer peak was identified to exist at the

wavenumber of 2237 cm<sup>-1</sup>. In addition, aromatic ring for stryrene was found at wavenumber 1643 cm<sup>-1</sup> and the double bond of butadiene and vynil was observed at 964 cm<sup>-1</sup> and 910 cm<sup>-1</sup>, respectively [27].



Figure 4. FTIR spectra of (a) kenaf fibers, (b) recycled ABS, (c) rABS-KSF-10%, and (d) rABS-KSF-15%.

FTIR spectra for the biocomposites were dominated by peaks of the ABS polymer. There were no significant transmittance changes at acrylonitrile monomer peak. Aromatic ring for styrene experiences the change of transmittance with the increase in the filler content in the biocomposites. The change for OH peak transmittance in the biocomposites indicated a reaction between the hydroxyl groups of filler binding with the carboxyl hydroxyl groups of maleic anhydride [28]. Maleic anhydride was successfully grafted on butadiene chain.

Based on the FTIR spectra, intensity of butadiene in the biocomposites was changed when it is compared with ABS polymer because the double bond of butadiene (C<sub>4</sub> and C<sub>5</sub>) binded C<sub>2</sub> and C<sub>3</sub> from maleic anhydride by covalent bonding. Figure 3 also shows changes in the transmittance intensity for trans and vynil, which are increased with increasing filler contents in the biocomposites. The wavenumber of ester carbonyl (C=O) was observed at 1728 cm<sup>-1</sup>, 1751 cm<sup>-1</sup>, 1790 cm<sup>-1</sup>, and 1736 cm<sup>-1</sup> for vABS-KSF-10%, vABS-KSF-15%, rABS-KSF-10%, and rABS-KSF-15%, respectively. This ester carbonyl promotes the binding between the kenaf fibers with maleic acid by esterification. Esterification of cellulose with carboxylic derivatives occurs through a mechanism of derivatives cyclic carboxylic acid due to heating treatment at a temperature of 170 °C [29].

Analysis of the functional groups on the biocomposites showed the shifting and changing of the functional groups that are influenced by their treatment procedure. These shifting and changing of the functional groups for the biocomposites also demonstrated variation in the chemical bonding which is possibly due to the changes of tissue structure of the polymer matrix. In general, findings from this study contribute significantly on the development of materials derived from natural resources. Although synthetic materials are also promising for certain applications [30-

# 4. CONCLUSION

The present study has discussed the fabrication and evaluation of the physio-chemical features of composites filled with natural resources. It was interesting to note that the addition of short kenaf fibers for fabricating biocomposites affected their density and the melt flow index. Surface free energy of the biocomposites fabricated using virgin ABS was demonstrated to decrease with

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increasing filler content. However, it was found that their characteristic of biocomposites fabricated using recycled ABS was slightly improved. In addition, FTIR observation exhibited that the addition of the natural fibers in the virgin and recycled ABS performed a slightly different characteristic.

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