

Rheological Properties of Injectable Hyaluronic Acid Hydrogels for Soft Tissue Engineering Applications

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Abstract: Hydrogels are cross-linked three-dimensional (3D) polymeric network, which can hold the water within its porous structure. They have recently been used in various biomedical applications. In this study, injectable hyaluronic acid (HA) hydrogels were prepared using different concentrations of 1,4-butanediol diglycidyl ether (BDDE) as a crosslinker (1-5% w/w) and investigated their rheological, swelling and injectability properties. The results demonstrated that the rheological characteristics of hydrogels enhanced with increasing crosslinker concentration. The elastic modulus of the hydrogels ranged from 280 Pa to 990 Pa, while the complex viscosities were found between 42 Pa.s and 190 Pa.s at an oscillation frequency of 1 Hz. These results clearly suggest that the injectable HA hydrogels are a potential candidate for various soft tissue engineering applications due to their highly tunable rheological properties.

Keywords: Crosslinking; biomaterials; hydrogel; viscoelasticity; rheology; soft tissue engineering.

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1. Introduction

Hydrogels are promising three-dimensional (3D) polymeric networks which have outstanding characteristics such as high water uptake, good biocompatibility, and tunable biomechanical and viscoelastic properties [1–4]. They have been mostly used in many biomedical applications, including drug delivery, wound dressing, biosensors, prevention of postoperative adhesions, orthopedic or aesthetic injections, and contact lenses [5–12]. In particular, injectable hydrogels have considerably attracted interest in the biomedical industry, due to their low invasive surgical procedure, adhering to tissue and filling properties [13–16].

Hyaluronic acid (HA) is one of the most preferred biopolymers in injectable hydrogels, which consists of repeating D-glucuronic acid and N-acetyl D-glucosamine disaccharide units [17–19]. It exhibits interesting viscoelastic and lubricating properties, excellent biocompatibility, and biodegradability [20]. In contrast to these advantages, its poor mechanical stability and rapid degradation limit its use in biomedical applications [21]. It is a known fact that the half-life of HA in the human body is less than one day [22]. Therefore, it cannot be expected to be long-lasting in injected areas. To overcome this drawback, many cross-linking agents such as 1,4-butanediol diglycidyl ether (BDDE), 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide (EDC), divinyl sulfone (DVS) and glutaraldehyde (GTA) have been used to improve its biomechanical properties [23–25]. Among them, BDDE is widely used in much dermal filler and intraarticular viscosupplements due to its low

cytotoxicity and sensitization. It can easily react with hydroxyl groups to form ether bonds by the epoxide ring-opening of itself in an alkaline environment [26].

The crosslinker ratio is strongly associated with the degree of cross-linking. The cross-linking degree is increased with the increase of crosslinker amount resulting in highly stable and durable hydrogel against enzymatic degradation. Therefore, in this study, injectable HA hydrogels were prepared at different concentrations of BDDE, and the effect of crosslinker concentration on rheological properties was evaluated.

2. Materials and Methods

2.1. Materials.

Hyaluronic acid sodium salt (molecular weight = 7.66×10^5 Da) was purchased from Shiseido. 1,4-butanediol diglycidyl ether (BDDE) was provided from Sigma Aldrich. All other chemicals were of analytical grade and used as received without any purification.

2.2. Preparation of HA hydrogels at different crosslinker concentrations.

10 wt % of HA sodium salt was first dissolved in 0.25 M sodium hydroxide (NaOH) solution by a mechanical stirrer in 300 rpm for 30 min. After the HA was completely dissolved, BDDE at different concentrations (1, 2, 3, and 5 wt % with respect to the total mixture) was added to HA solution and stirred for an additional 5 min. The cross-link reaction was then performed at 45 °C for 4 hours. The prepared hydrogels were named as HA1, HA2, HA3, and HA5, respectively (Figure 1). After the reaction, the hydrogels were put into deionized water (DIW) following by neutralization with 0.1 M of hydrochloric acid (HCl). The hydrogels were then repeatedly washed and swollen in DIW to remove unreacted substances. The resulting hydrogels were then precipitated in excess ethanol and dried under vacuum oven at room temperature for 24 hours. Finally, the hydrogels with a concentration of 20 mg/ml were filled into 1 mL syringes and autoclaved in air-stream at 121 °C for 15 min.

2.3. Swelling Test.

A certain amount of dried hydrogels was immersed into a physiological saline solution (NaCl 0.9 % w/w) for 24 h and then separated by filtration. The swelling ratio (SR) of the hydrogels was calculated using Eq. (1). Measurements were performed three times, and the averages were calculated:

$$SR (g/g) = \frac{W_w}{W_d} \quad \text{Eq. (1)}$$

wherein W_w and W_d represent swollen and dry weights of the hydrogels, respectively.

2.4. Characterization.

Rheological measurements were performed using an Anton Paar MCR 102 rheometer equipped with a parallel plate geometry, 25 mm plate diameter and 1.0 mm gap, while the injection force tests were carried out at a compression rate of 12 mm/min by Lloyd-LS1 testing device.

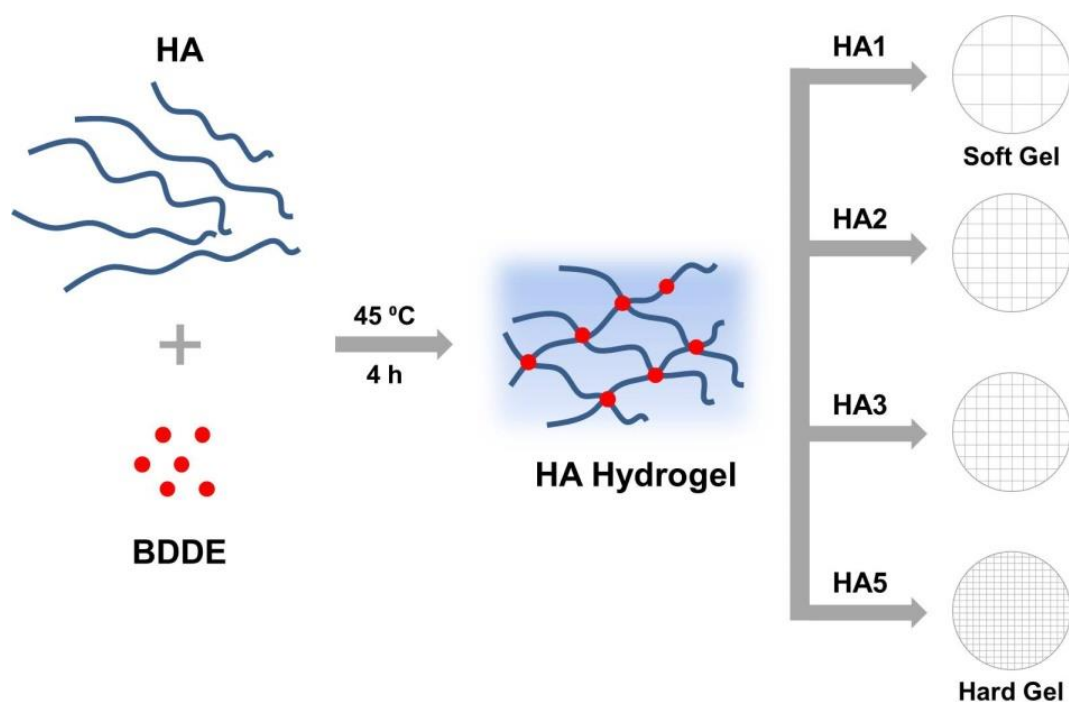


Figure 1. Schematic illustration of HA hydrogels at different crosslinker concentration.

3. Results and Discussion

The swelling ability of the hydrogels is a crucial parameter because it characterizes their capacity to maintain the structure and retain sufficient water, affecting all characteristics of the hydrogel in relevant applications such as stiffness, degradation, and biocompatibility [27,28]. To understand the influence of the crosslinker concentration on the swelling behavior of the hydrogels, the swelling ratio was investigated by immersing the dried hydrogels in saline solution to reach equilibrium. As shown in Figure 2a, all hydrogels showed various swelling ratios, ranging from 42 to 194 g/g. The water uptake ability significantly decreased, as the crosslinker concentration increased from 1 to 5 wt %. These findings indicated that the hydrogels prepared at different crosslinker concentrations might be dominant on swelling behavior.

Injectable hydrogels with different biomechanical properties possess different rheological characteristics such as viscosity, elasticity, and plasticity. The rheological properties of injectable hydrogels are clinically relevant because they play a critical role in identifying how the hydrogel behaves after injection. The most significant rheological parameters for injectable hydrogels are regarded as complex viscosity (η^*) and elastic modulus (G'). Complex viscosity measures the ability of the gel to resist shearing forces within a tissue after injection, while elastic modulus measures the stiffness of hydrogel and its interactions [29]. These parameters determine how well it resists tension forces due to injected area movements after injection.

The elastic modulus of hydrogels as a function of the oscillation frequency are shown in Figure 2b. It can be noted that the elastic modulus of hydrogels increased with increasing crosslinker concentration. The elastic modulus of HA1, HA2, HA3, and HA5 were found as 280 Pa, 450 Pa, 680 Pa, and 990 Pa at 1 Hz, respectively (Figure 2c). The elastic modulus is especially important in determining the place of use in dermal filler applications. The high elastic modulus shows that the filler is firm and should be placed in deeper facial areas to reduce the tangibility in the dermis [30]. These types of highly resistant fillers can also be used

to correct nasal bridges. The fillers with lower elastic modulus show more fluid character, and they can be used in fine lines, which is more sensitive and over larger areas like the cheeks or cheekbones. Consequently, the results exhibited that the rheological properties of HA hydrogels improved with increasing crosslinker concentration. In addition, it has been shown that hydrogels with different elastic modulus values can be used as dermal fillers in different areas of the face.

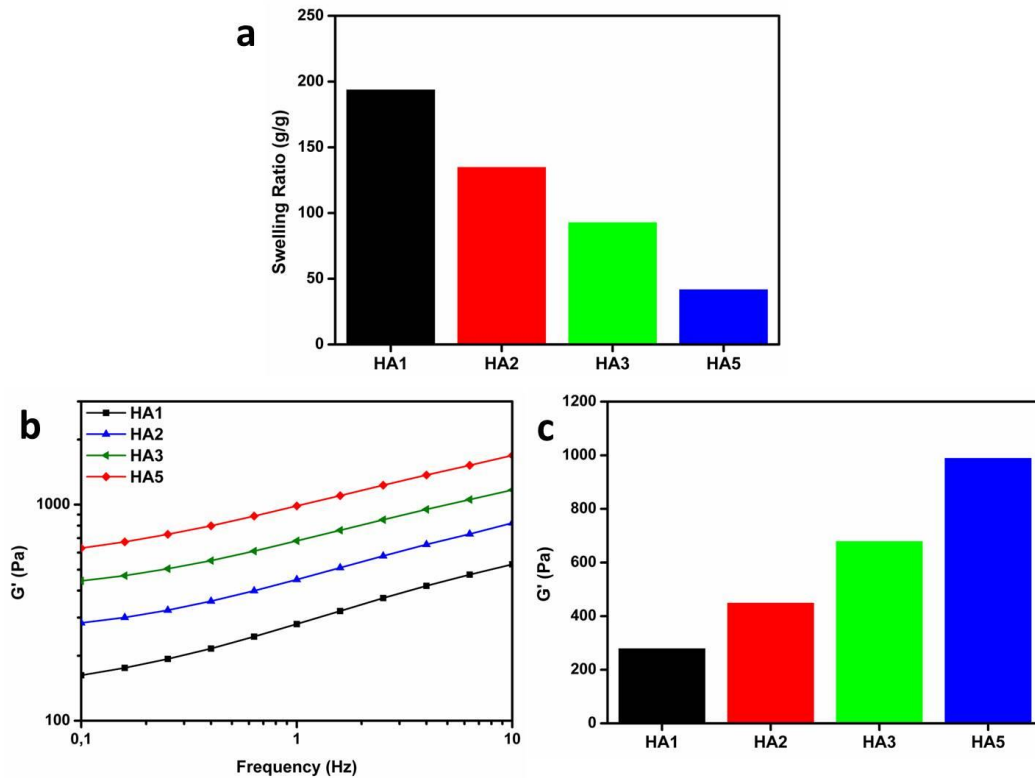


Figure 2. a) Swelling ratios of HA hydrogels, b) Elastic (G') modulus of hydrogels as a function of the oscillation frequency, c) bar graph showing G' modulus for hydrogels.

On the other hand, the complex viscosity at 1 Hz of hydrogels was shown in Figure 3a. The complex viscosity greatly decreased with increasing oscillation frequency, demonstrating that the hydrogels were highly shear thinning. The differences in complex viscosity can be related to the structural strength of hydrogels having different crosslinker concentrations. As shown in Figure 3b, hydrogels exhibited a higher viscosity with increasing crosslinker concentration. The complex viscosities were found as 42, 74, 107, and 190 Pa.s for HA1, HA2, HA3, and HA5, respectively. This showed that the complex viscosity of the hydrogels was influenced by the crosslinker concentration, and the increase in the crosslinker content of the hydrogel resulted in high complex viscosity. Moreover, injection force tests were performed to demonstrate the injectability of hydrogels during application. As shown in Figure 3c, it was clear that the injection force increased from 10 N to 31 N when the crosslinker concentration increased from 1 wt % to 5 wt % in the hydrogel matrix. Indeed, the injection forces in this regime are proportional to the elastic modulus. This was related to the high viscosity of the hydrogels when a higher crosslinker concentration was introduced. The results demonstrated that the increase in crosslinker concentration increased the injection force. When considering all factors, the increase in crosslinker concentration improves the rheological properties of the hydrogels but also leads to the injection force.

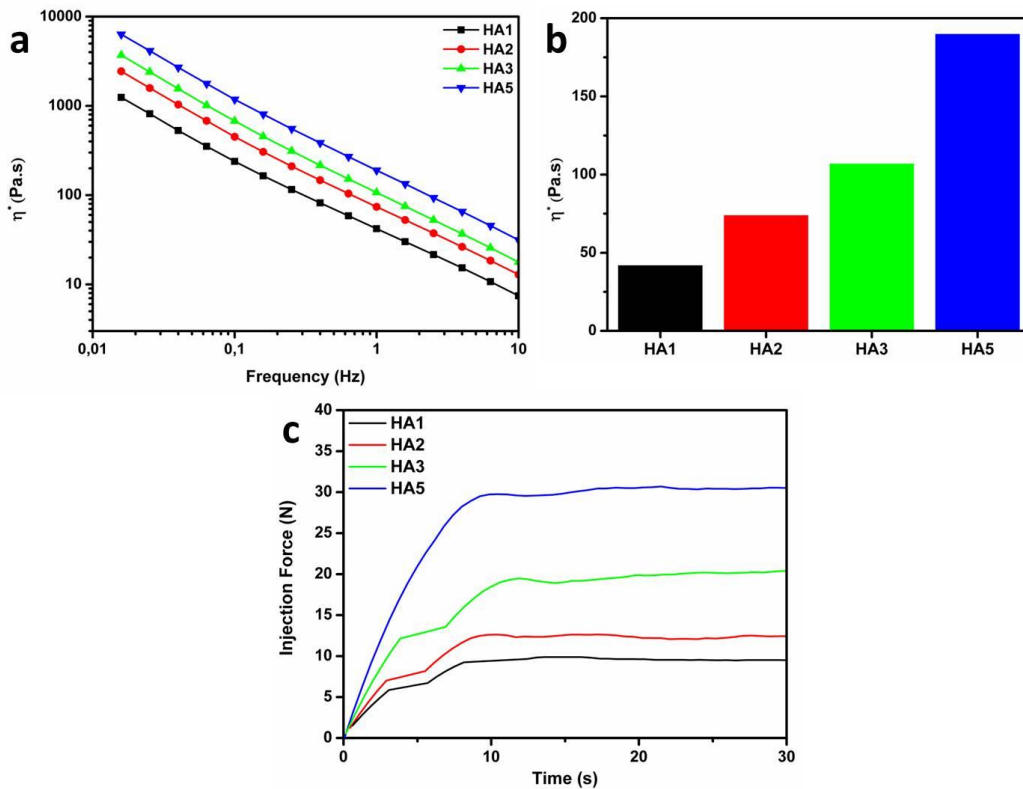


Figure 3. a) Complex viscosity (η^*) as a function of the oscillation frequency, b) bar graph showing complex viscosity for hydrogels, c) Graph showing extrusion force overtime for hydrogels injected from a 1 mL syringe with needles of 27G.

4. Conclusions

In this study, HA hydrogels were prepared at different concentrations of BDDE, and the effect of crosslinker concentration on rheological properties was evaluated. With increasing crosslinker concentration in the hydrogel matrix, the swelling ability of hydrogels significantly decreased, while they exhibited high elastic modulus and complex viscosity. The rheological analyses showed that the elastic modulus and complex viscosity of hydrogels were strongly dependent on crosslinker concentration, offering a means of controlling the biomechanical properties for injectable applications. The results clearly suggest that the HA hydrogels are especially a superior candidate for injectable applications due to their highly tunable properties.

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Conflicts of Interest

The authors declare no conflict of interest,

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