

# Response Surface Optimization and Impact of Immobilized Enzymes Naringinase and Tannase on the Quality Parameters of *Citrus maxima* Juice

Sanjay Kumar <sup>1,\*</sup> , Vijay Kumar <sup>2</sup> , Pankaj Gautam <sup>1</sup>

<sup>1</sup> Department of Life Sciences (Food Technology), Graphic Era Deemed to Be University, Dehradun

<sup>2</sup> Himalayan School of Biosciences, Swami Ram Himalayan University, Jolly Grant, Dehradun

\* Correspondence: [sanjaykumar@geu.ac.in](mailto:sanjaykumar@geu.ac.in);

Scopus Author ID 57208487301

Received: 29.10.2020; Revised: 2.12.2020; Accepted: 8.12.2020; Published: 12.12.2020

**Abstract:** Pomelo has been reported as a rich source of flavanone glycoside with antioxidants and exhibits favorable health effects such as antimicrobial, anti-inflammatory, antiatherogenic, antitumor, and anti-clotting activity. Despite all the beneficial health impacts of *Citrus maxima*, it still has lower commercial value because of its juice's bitterness due to the presence of naringin and tannic acid. Therefore, an attempt has been made for the cost-effective and economic debittering process using naringinase and tannase enzymes. The 17 experiments were planned according to RSM, BBD to analyze the effect of independent variables with three levels of each, i.e., Enzyme ratio ((Naringinase: tannase) (100:0, 50:50, 0:100)), incubation temperature (30°C, 40°C, 50°C) and incubation time (2, 3, 4 hrs) on physicochemical quality of *Citrus maxima* Juice. The study's result indicated that independent variables affected the responses (pH, TSS, TA, Naringin content, Tannin content, TPC, and Vitamin C content). Optimization was done using Design Expert 10.0.1 software to debitter and clarify *citrus maxima* juice by immobilized enzymes. The optimum values were found to be 54.55, 50°C, and 4 hrs. The values for pH, TSS, TA, Naringin content, Tannin content, TPC and Vitamin C content were found to be 3.17, 6.256 °Brix, 0.885 %citric acid, 220.549 µg/ml, 0.311mg/ml, 1256.721 mg GAE/L, 30.309 mg/100ml respectively. From the study, it could be concluded that the maximum debittering and clarification of *citrus maxima* juice could be done under processing conditions, i.e., enzyme ratio 50:50, incubation temperature 50°C and incubation time 4 hrs.

**Keywords:** *Citrus maxima*; naringinase; tannase; total phenolic content; tannin content; naringin content.

© 2020 by the authors. This article is an open-access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (<https://creativecommons.org/licenses/by/4.0/>).

## 1. Introduction

*Citrus maxima* (or *Citrus grandis*), commonly known as Chakotra, pomelo, and jabong [1] has been reported as a rich source of naringin, a bitter flavored, flavanone glycoside with antioxidant [2], and exhibit favorable health effects such as antimicrobial, anti-inflammatory, antiatherogenic, antitumor, and anti-clotting activity [3-5]. *Citrus maxima* juice contains anti-nutrients like phytic acid, tannin, and oxalate [6]. Bitter compounds are present in a different part of a single fruit [7]. Naringin is the source of undesirable bitterness [8], especially in the citrus fruit juice industry; therefore, it must be removed or reduced from processed products [9-11]. Also, upon storage of the juices, due to tannin content, certain factors such as sedimentation, haziness, color, astringency [12-13], and bitterness get increased [14-15]. These problems can be solved using the enzymes naringinase produced by different microbial strains

[16] and tannase produced by several microbial strains [17]. Naringinase hydrolyzes naringin to naringenin, a non-bitter derivative, which cannot be reconverted to naringin [18-19] and glucose [20-21], resulting in an improvement in the taste of citrus juice [22]. Besides, naringinase plays an important role in modifying flavonoids to yield highly bioactive compounds [23-24]. Similarly, tannin acyl hydrolase, i.e., tannase enzyme (EC 3.1.1.20), has been used to produce gallic acid and glucose from tannins [25]. This enzyme has found its role in various applications world-wide, especially in pharmaceuticals, tannery, beverage, and alcohol industries for clarification purposes [26-27].

Despite all the beneficial health impacts of *Citrus maxima*, it still has lower commercial value because of its juice's bitterness. Therefore, a cost-effective and economic debittering process could be achieved if naringinase and tannase produced industrially using microorganisms [28]. Hence, this study's objective was debittering and clarifying *Citrus maxima* juice by immobilized enzyme naringinase and tannase produced from *Aspergillus sp.* isolate SK01 isolated from rotten *Citrus maxima*.

## 2. Materials and Methods

### 2.1. Materials.

*Citrus maxima* of sound quality were brought from Dehradun's local market and washed properly 2-3 times with tap water and then with distilled water to remove dirt. The flavedo and albedo were peeled and separated from the pulp. The pulp was blended and filtered through a muslin cloth to obtain a clear juice and stored in a refrigerator at 4°C for further analysis. All the chemicals of the analytical grade used in this study were purchased from Hi-Media, Mumbai, India.

### 2.2. Production of enzymes.

Enzymes naringinase and tannase were produced through submerged fermentation by the method of [29-30] and [31] respectively from *Aspergillus sp.* isolate SK01 and partially purified by ammonium sulfate (80%) ppt method. Naringinase activity was assayed with respect to naringin using Davis's method [32] with little modification. Tannase activity was determined calorimetrically using the method of Mondal [33]. The activity of partially purified crude enzymes naringinase and tannase was 1.63 IU/ml and 1.18 IU/ml, respectively.

### 2.3. Immobilization of naringinase.

Immobilization of Naringinase was done by [34] with some modifications. Briefly, a 5 ml suspension containing crude naringinase at a concentration of 1 g/l and 3% sodium alginate was extruded into a 0.2 M CaCl<sub>2</sub> solution at a temperature of 4°C to form the gel beads. After 4 h the beads were washed 2-3 times with 0.1 M sodium acetate buffer, pH 4.0, and used for naringinase assay.

### 2.4. Immobilization of tannase.

Tannase was done by [35] with some modifications. Briefly, a 5 ml suspension containing tannase at a concentration of 1 g/l and 3% sodium alginate was extruded into a 0.2 M CaCl<sub>2</sub> solution at a temperature of 4°C to form the gel beads. After 4 h the beads were washed 2-3 times with water and kept at 4°C.

### 2.5. Identification and selection of the most important variables.

Several experiments were carried out to standardize the parameters for debittering and clarification of *Citrus maxima* juice by immobilized enzyme naringinase and tannase produced from *Aspergillus* sp. isolate SK01 isolated from rotten *Citrus maxima*.

Enzyme ratios were selected to show the effect of naringinase and tannase individually or in a combination of both enzymes on debittering and clarification characteristics of *Citrus maxima* juice as an individual or in a combination of both. The enzyme ratios range (100:0, 50:50, 0:100) were selected. Based on preliminary experiments, incubation temperature (30°C, 40°C, 50°C) and incubation time (2, 3, 4 hrs) were selected as variables.

### 2.6. Physiochemical analysis.

pH was measured by handy pH meter (Eutech), TSS was measured by using a hand refractometer (ERMA). Titrable acidity was measured by [36], Vitamin C (mg/100ml) was measured by the method of [37-38]. Total Phenolic Content (mg GAE/L) was measured as described [39]. Tannin content (mg/ml) and naringin content (µg/ml) in *Citrus maxima* juice were determined by the method of [40- 41], respectively.

### 2.7. Statistical analysis.

Design-Expert 10.0.1 was used for data analysis and process optimization. To evaluate the effect of process parameters, i.e., enzyme ratios (naringinase: tannase), incubation temperature (°C) and incubation time (hrs.) on the responses, i.e., pH, TSS, titrable acidity, naringin content, tannin content, TPC and vitamin C content a second-order response function was implemented for three independent variables having following general form Equation (1) [42-43].

$$Y = \beta_0 + \sum_{i=1}^3 \beta_i X_i + \sum_{i=1}^2 \sum_{j=i+1}^3 \beta_{ij} X_i X_j + \sum_{i=1}^3 \beta_{ii} X_i^2 \quad (1)$$

Where,

Y= Response

$\beta_0$ ,  $\beta_{ii}$ ,  $\beta_{ij}$  = Coefficients

$X_i$  and  $X_j$  = Independent Variables

ANOVA was used for determining the statistical significance of the independent parameters and their relative interactions. The model's adequacy was explained in terms of  $R^2$  (coefficient of determination), F-value (Fisher's value), and LOF (lack-of-fit)

## 3. Results and Discussion

A total of 17 experiments were designed by using the Box-Benkhen design of RSM. These runs were performed to see the effect of the selected process parameters- enzyme ratio, temperature & time on the said responses, i.e., naringin content, tannin content, TPC, and vitamin C. The results were statistically analyzed for being either significant or non-significant. The results of the experiment are given in Table 1. Table 2 and Table 3 show the ANOVA and regression analysis performed to check the model's adequacy. Constraints for optimization for independent variables/ dependent variables are given in Table 5.

### 3.1 Effect on pH

Full second order equation to show the effect of  $X_1$ ,  $X_2$  and  $X_3$  on pH could be explained by the equation given below.

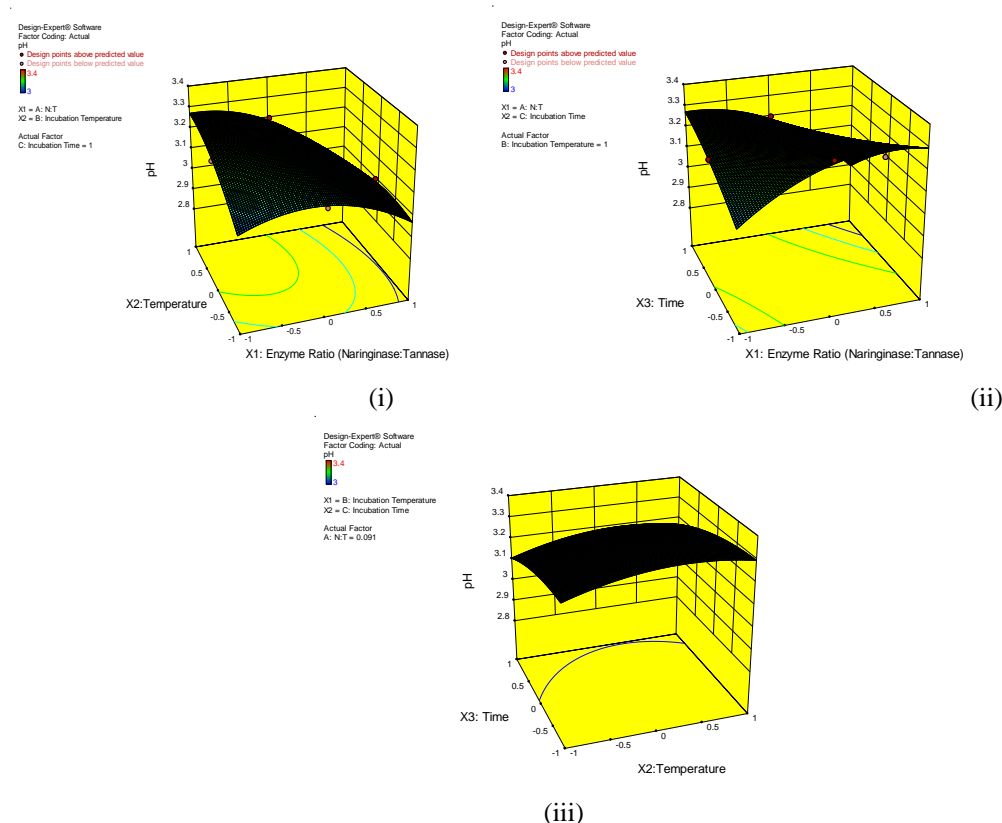
$$\text{pH} = 3.28 + 0.0375 X_1 + 0.0375 X_2 - 0.050 X_3 - 0.075 X_1 X_2 - 0.15 X_1 X_3 + 1.08 \times 10^{-16} X_2 X_3 - 0.10250 X_1^2 - 0.0525 X_2^2 - 0.0275 X_3^2$$

Where,

$X_1$ ,  $X_2$  and  $X_3$  are Enzyme ratios (naringinase: tannase), temperature, and time respectively.

The total effect of individual parameters on pH at linear, quadratic, and interactive levels is reported in table 3. The F value (16.22) for the model was significant at 1% ( $p < 0.01$ ) level of. Correlation coefficient  $R^2$  measures the generosity of fit of the model. The  $R^2$  value for pH was 95.42%, which means that the model could account for 95.42% of the data. The model does not elucidate the variation of 4.58%. The  $R^2$  value higher than 90% showed that the regression model explained the reaction well. The Pred R-Squared of 77.12% was in equitable agreement with the Adj R-Squared of 89.54 %. Lack of fit, i.e., LOF was insignificant; hence, the second-order model was acceptable in describing pH. The pH value for debittered juice was varied from 3.0 to 3.4 (Table 3). The highest pH score was found maximum (3.4) at the level of  $X_1$ (0:100),  $X_2$  (40°C), and  $X_3$ (2 hrs).

The pH value for debittered juice was varied from 3.0 to 3.4 (Table 3). The highest pH score was found maximum (3.4) at the level of  $X_1$ (0:100),  $X_2$  (40°C), and  $X_3$ (2 hrs).



**Figure 1.** 3D Response surface showing interaction effect of variables on pH of debittered *Citrus maxima* juice i) temperature and enzyme ratio ii) time and enzyme ratio and iii) time and temperature.

The 3D response surface curve for pH was presented in Figure 1. (i, ii & iii). The values presented in Table 3 show that there were no significant changes in the pH of the debittered juice. Figure 1 (i) shows that as temperature increase (30°C to 50°C), pH was increased in the presence of enzyme ratio.

**Table 1.** Experimental design.

Expt. No.	Coded Levels			Real Levels		
	X <sub>1</sub>	X <sub>2</sub>	X <sub>3</sub>	Enzymes Blend ratios (Naringinase: Tannase)	Temperature (°C)	Time (hrs)
1	-1	-1	0	100:0	30	3
2	1	-1	0	0:100	30	3
3	-1	1	0	100:0	50	3
4	1	1	0	0:100	50	3
5	-1	0	-1	100:0	40	2
6	1	0	-1	0:100	40	2
7	-1	0	1	100:0	40	4
8	1	0	1	0:100	40	4
9	0	-1	-1	50:50	30	2
10	0	1	-1	50:50	50	2
11	0	-1	1	50:50	30	4
12	0	1	1	50:50	50	4
13	0	0	0	50:50	40	3
14	0	0	0	50:50	40	3
15	0	0	0	50:50	40	3
16	0	0	0	50:50	40	3
17	0	0	0	50:50	40	3

Figure 1 (ii) showed that the pH increase as time increases (2-4 hrs) with different enzyme ratio levels. Figure 1. (iii) depicted that the pH increased along with an increasing level of time and temperature. A similar finding was observed by[44] in the case of the enzymatic treatment of pomegranate juice.

From table 6 of coefficient, it was observed that X<sub>1</sub> (enzyme ratio) and X<sub>2</sub> (temperature) had a positive effect on pH at 5% level, i.e., p<0.05 of significance at linear level, while X<sub>3</sub> (Time) had a negative effect at 1% level, i.e., p<0.01 of significance. Interactive effect of X<sub>2</sub> (temperature), X<sub>3</sub> (time) had significant positive effect at 10% level i.e. p<0.1 of significance while X<sub>1</sub> (enzyme ratio), X<sub>2</sub> (temperature) and X<sub>1</sub> (enzyme ratio), X<sub>3</sub> (time) had negative effect at 1% level i.e. p<0.01 of significance. Quadratic effect of X<sub>1</sub> (enzyme ratio), X<sub>2</sub> (temperature) and X<sub>3</sub> (time) was negative at 1% i.e. p<0.01, 5% level i.e. p<0.05 and more than 10% level i.e. p<0.1 of significance respectively.

**Table 2.** Independent variables (coded and actual value).

Independent variables		Coded Levels		
Name	Code	-1	0	1
Actual Levels				
Enzymes Blend ratios (Naringinase: Tannase)	X <sub>1</sub>	100:0	50:50	0:100
Temperature (°C)	X <sub>2</sub>	30	40	50
Time (hrs)	X <sub>3</sub>	2	3	4

### 3.2. Effect on TSS.

The effect of X<sub>1</sub>, X<sub>2</sub>, and X<sub>3</sub> on TSS could be explained by the equation given below.

$$\text{TSS} = 7.26 - 1.1625 X_1 - 0.0375 X_2 - 0.0250 X_3 + 0.075 X_1 X_2 + 0.150 X_1 X_3 - 0.200 X_2 X_3 + 0.482 X_1^2 - 0.217 X_2^2 - 0.44250 X_3^2$$

**Table 3.** Experimental data for debittering and clarification of pomelo juice by using immobilized enzyme naringinase and tannase and combination of both.

Variables				Responses						
Exp. No.	Enzymes Blend ratios (Naringinase: Tannase)	Temp. (°C)	Time (hrs)	pH	TSS (°Brix)	TA (% citric acid)	Naringin content (µg/ml)	Tannin content (mg/ml)	TPC mgGAE/L	Vitamin C content (mg/100ml)
1	100:0	30	3	3.0*	8.8**	0.98	505.8	0.68**	1819.8	58.2**
2	0:100	30	3	3.2	6.3	0.97	686.5**	0.63	1632.8	55.2
3	100:0	50	3	3.2	8.6	0.99**	213.8	0.60	1822.8**	52.2
4	0:100	50	3	3.1	6.4	0.92	544.3	0.41	1518.8	47.6
5	100:0	40	2	3.0*	8.7	0.85	419.5	0.65	1660.7	53.4
6	0:100	40	2	3.4**	6.1*	0.84*	659.8	0.62	1177.4	53.1
7	100:0	40	4	3.2	8.2	0.86	414.2	0.68**	1695.7	54.4
8	0:100	40	4	3.0*	6.2	0.85	612.4	0.55	1362.7	45.4
9	50:50	30	2	3.2	6.4	0.89	415.5	0.52	1218.6	44.5
10	50:50	50	2	3.3	6.7	0.88	226.5	0.39	1230.3	37.5
11	50:50	30	4	3.1	6.9	0.91	354.3	0.55	1389.6	38.8
12	50:50	50	4	3.2	6.4	0.89	216.9*	0.31*	1265.7	29.3
13	50:50	40	3	3.3	7.3	0.97	365.5	0.50	1125.3	26.8
14	50:50	40	3	3.3	7.3	0.95	304.2	0.49	1048.0*	24.8
15	50:50	40	3	3.3	7.3	0.96	305.3	0.50	1125.3	24.8
16	50:50	40	3	3.3	7.2	0.95	305.5	0.50	1124.3	24.5*
17	50:50	40	3	3.2	7.2	0.97	363.5	0.48	1125.3	26.8

\*Minimum value

\*\*Maximum value

**Table 4.** ANOVA for different responses.

Source	Degree of freedom	pH		TSS (°Brix)		TA (% citric acid)		Naringin content (µg/ml)		Tannin content (mg/ml)		TPC (mgGAE/L)		Vitamin C content (mg/100ml)	
		F value	P Value Prob>F	F value	P Value Prob>F	F value	P Value Prob>F	F value	P Value Prob>F	F value	P Value Prob>F	F value	P Value Prob>F	F value	P Value Prob>F
Model	9	16.22	0.0007	170.08	< 0.0001	37.17	< 0.0001	42.54	< 0.0001	76.04	< 0.0001	47.76	< 0.0001	147.92	< 0.0001
X <sub>1</sub>	1	7.50	0.0290	1271.91	< 0.0001	10.00	0.0159	123.20	< 0.0001	80.23	< 0.0001	83.07	< 0.0001	18.87	0.0034
X <sub>2</sub>	1	7.50	0.0290	1.32	0.2877	4.90	0.0625	79.03	< 0.0001	225.09	< 0.0001	2.42	0.1636	59.86	0.0001
X <sub>3</sub>	1	13.33	0.0082	0.59	0.4682	2.50	0.1579	2.08	0.1921	4.06	0.0837	8.85	0.0207	28.04	0.0011
X <sub>1</sub> X <sub>2</sub>	1	15.00	0.0061	2.65	0.1478	7.20	0.0314	6.13	0.0425	19.66	0.0030	1.33	0.2865	0.34	0.5791
X <sub>1</sub> X <sub>3</sub>	1	60.00	0.0001	10.59	0.0140	0.000	1.0000	0.48	0.5090	10.03	0.0158	2.20	0.1819	10.00	0.0159
X <sub>2</sub> X <sub>3</sub>	1	0.000	1.0000	18.82	0.0034	0.20	0.6682	0.73	0.4219	12.13	0.0102	1.79	0.2230	0.83	0.3937
X <sub>1</sub> <sup>2</sup>	1	29.49	0.0010	115.32	< 0.0001	11.84	0.0108	167.84	< 0.0001	304.42	< 0.0001	253.51	< 0.0001	972.52	< 0.0001
X <sub>2</sub> <sup>2</sup>	1	7.74	0.0272	23.43	0.0019	19.00	0.0033	4.77	0.0653	39.32	0.0004	62.52	< 0.0001	104.57	< 0.0001
X <sub>3</sub> <sup>2</sup>	1	2.12	0.1885	96.99	< 0.0001	280.47	< 0.0001	0.21	0.6638	0.18	0.6854	1.38	0.2789	58.56	0.0001
Lack of fit	3	0.42	0.7510	5.28	0.0709	1.58	0.3258	0.68	0.6108	5.94	0.0591	3.72	0.1184	1.97	0.2612
Std. Dev.		0.039		0.092		0.011		30.25		0.016		50.71		1.38	
Mean		3.19		7.18		0.92		406.68		0.53		1373.12		41.02	
R <sup>2</sup>		0.9542		0.9954		0.9795		0.9820		0.9899		0.9840		0.9948	

$R^2_{adj}$		0.8954		0.9896		0.9532		0.9590		0.9769		0.9634		0.9880	
<b>Pred R- Squared</b>		0.7712		0.9404		0.8074		0.8848		0.8648		0.8047		0.9468	
<b>Adeq Precision</b>		13.887		39.421		17.784		20.062		30.451		19.377		31.503	
<b>CV (%)</b>		1.21		1.28		1.22		7.44		2.96		3.69		3.55	
<b>Model</b>		Sig		Sig		Sig		Sig		Sig		Sig		Sig	
<b>Lack of Fit</b>		NOT Sig		NOT Sig		NOT Sig		NOT Sig		NOT Sig		NOT Sig		NOT Sig	

**Table 5.** Constraints for optimization for independent variables/ dependent variables by design Expert 10.0.1.

Constraints Name	Goal	Lower Limit	Upper Limit	Lower Weight	Upper Weight	Importance
X <sub>1</sub> :Enzyme Ratio (Naringinase:Tannase)	is in range	-1	1	1	1	3
X <sub>2</sub> : Temperature (°C)	is in range	-1	1	1	1	3
X <sub>3</sub> : Time (hrs)	is in range	-1	1	1	1	3
pH	is in range	3	3.4	1	1	3
TSS (°Brix)	minimize	6.1	8.8	1	1	3
TA (% citric acid)	is in range	0.84	0.99	1	1	3
Naringine content (µg/ml)	minimize	213.8	686.5	1	1	5
Tannin content (mg/ml)	minimize	0.31	0.68	1	1	4
TPC (mgGAE/L)	is in range	1048	1822.8	1	1	3
Vitamin C Content (mg/100ml)	is in range	24.5	58.2	1	1	3

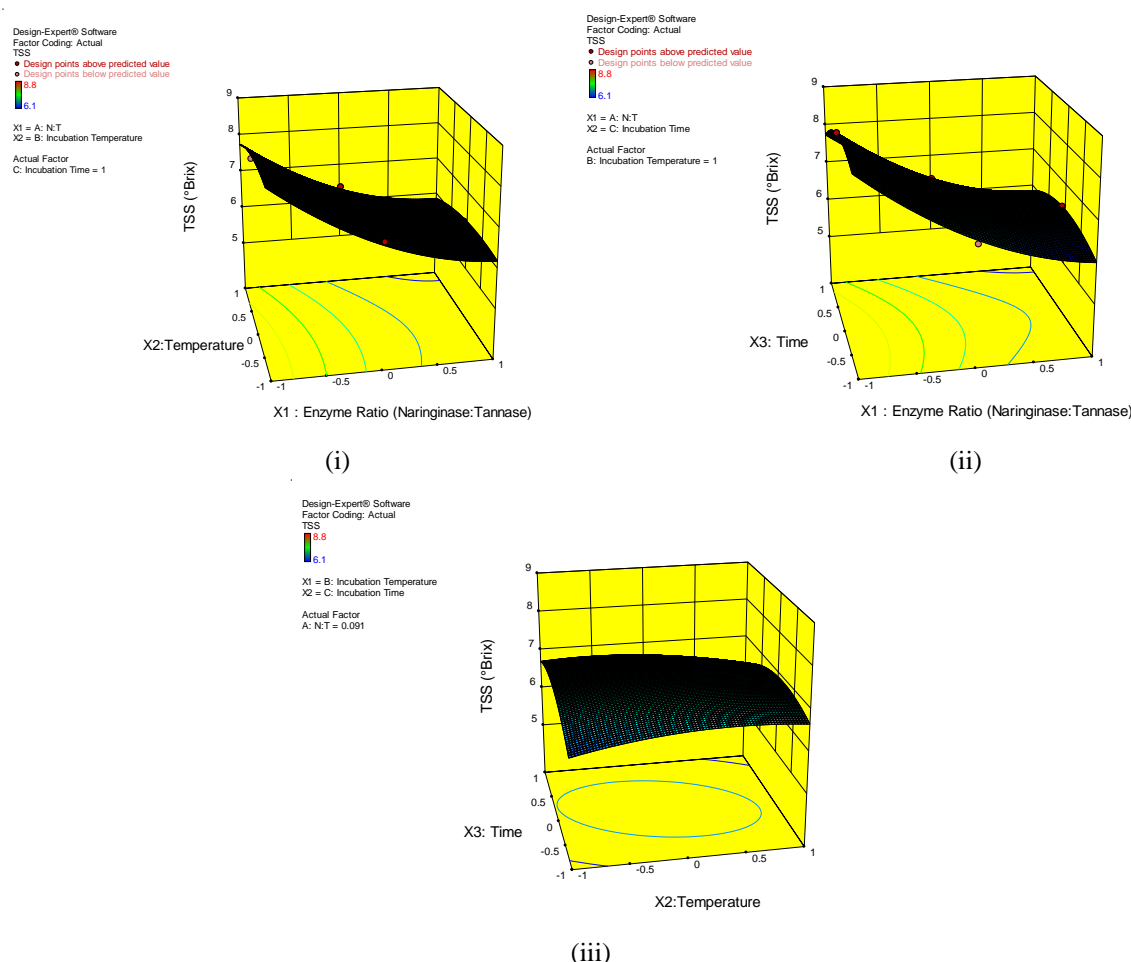
**Table 6.** Coefficients for different responses by design Expert 10.0.1.

Factor	Coefficients						
	pH	TSS °Brix	TA %	Naringine Content (µg/ml)	Tannin Content (mg/ml)	TPC (mgGAE/L)	Vit C Content (mg/100ml)
<b>Intercept</b>	3.28	7.26	0.96	328.8	0.494	1109.64	25.54
<b>X<sub>1</sub></b> <b>Enzyme Ratio (Nringinase: Tannase)</b>	0.0375**	-1.162***	-0.012**	118.713***	-0.05***	-163.412***	-2.1125***
<b>X<sub>2</sub></b> <b>(Incubation Temperature)</b>	0.0375**	-0.037	-0.0087*	-95.075***	-0.08375***	-27.9	-3.7625***
<b>X<sub>3</sub></b> <b>(Incubation Time)</b>	-0.05***	-0.025	0.0062	-15.4375	-0.01125*	53.3375**	-2.575***
<b>X<sub>1</sub>X<sub>2</sub></b>	-0.075***	0.075	-0.015**	37.45**	-0.035***	-29.25	-0.4
<b>X<sub>1</sub>X<sub>3</sub></b>	-0.15***	0.15**	-1.124 X 10 <sup>-18</sup>	-10.525	-0.025**	37.575	-2.175**
<b>X<sub>2</sub>X<sub>3</sub></b>	1.088 X 10 <sup>-16</sup>	-0.2***	-0.0025	12.9	-0.0275**	-33.9	-0.625
<b>X<sub>1</sub><sup>2</sup></b>	-0.1025***	0.4825***	-0.0187**	190.988***	0.13425***	393.492***	20.905***
<b>X<sub>2</sub><sup>2</sup></b>	-0.0525**	-0.217***	0.02375***	-32.1875*	-0.04825***	195.417***	6.855***
<b>X<sub>3</sub><sup>2</sup></b>	-0.0275	-0.442***	-0.09125***	6.6875	-0.00325	-29.0075	5.13***

\*\*\*, \*\*, \* Significant at 1,5and 10% level of significance respectively



The F value (170.08) for the model was significant at 1%, i.e.,  $p < 0.01$  level of significance. Correlation coefficient R measures the generosity of fit of the model.  $R^2$  value for TSS was 99.54%, which means that the model could account for 99.54% of data. The model does not elucidate 0.46% variation. The Pred R-Squared of 94.04 % was in equitable agreement with the Adj R-Squared of 98.96 %. LOF was insignificant; hence, the second-order model was acceptable in describing TSS. The variation in TSS value for debittered juice was observed from 6.1 to 8.8 (table 3). The TSS score was found minimum (6.1) at the level of  $X_1(0:100)$ ,  $X_2(40^\circ\text{C})$ , and  $X_3(2\text{ hrs})$ .



**Figure 2.** 3D Response surface showing interaction effect of variables on TSS of debittered *Citrus maxima* juice i) temperature and enzyme ratio ii) time and enzyme ratio and iii) time and temperature.

The 3D response surface curve for TSS was presented in figure. 2 (i, ii & iii). The values presented in Table 3 show that there were significant changes in TSS of the debittered juice (6.1-8.8). Figure 2 (i) shows that as temperature increase (30°C to 50°C), TSS was decreased in the presence of enzyme ratio. Figure 2 (ii) showed that the TSS decreased as time increased (2-4 hrs) with different enzyme ratio levels. Figure 2 (iii) depicted that the TSS increased along with an increasing level of time and temperature. Our finding favors the finding of [45].

From table 6 of the coefficient, it was observed that  $X_1$  (enzyme ratio) at 1% level, i.e.,  $p < 0.01$  of significance,  $X_2$  (temperature), and  $X_3$  (time) had a negative effect on TSS at a linear level. Interactive effect of  $X_2$  (temperature),  $X_3$  (time) had significant negative effect at 1% level i.e.  $p < 0.01$  of significance while  $X_1$  (enzyme ratio),  $X_2$  (temperature) had positive effect at more than 10% level ( $p < 0.1$ ) of significance whereas  $X_1$  (enzyme ratio),  $X_3$  (time) had positive effect at 5% level ( $p < 0.05$ ) of significance. Quadratic effect of  $X_1$  (enzyme ratio) was



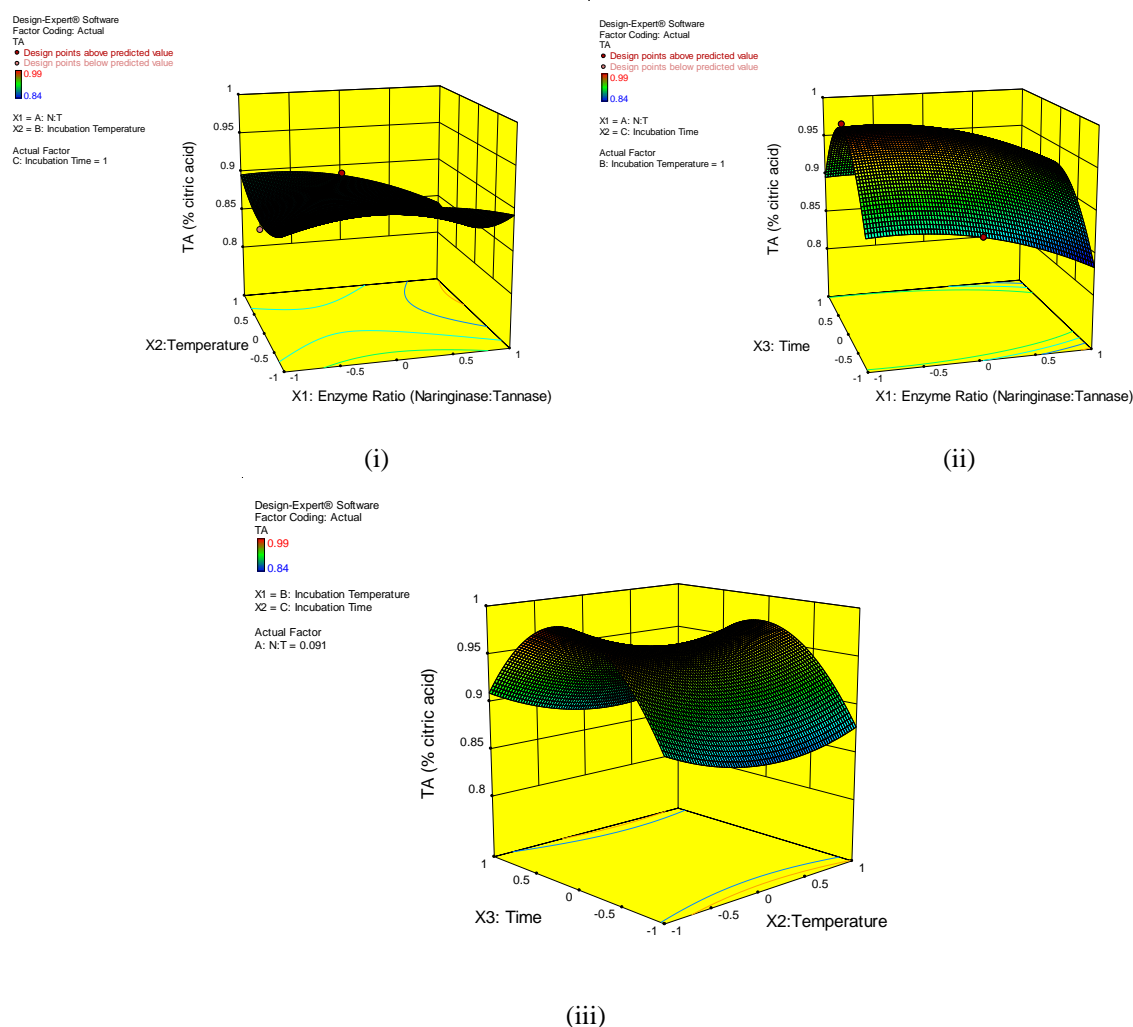
positive at 1% level ( $p < 0.01$ ) of significance, while  $X_2$  (temperature) and  $X_3$  (time) was negative at 1 level % ( $p < 0.01$ ).

### 3.3. Effect on TA.

Full second order equation to show the effect of  $X_1$ ,  $X_2$ , and  $X_3$  on TA could be explained by the equation given below.

$$TA = 0.96 - 0.012 X_1 - 8.75 \times 10^{-3} X_2 + 6.250 \times 10^{-3} X_3 - 0.015 X_1 X_2 - 1.124 \times 10^{-18} X_1 X_3 - 2.50 \times 10^{-3} X_2 X_3 - 0.018750 X_1^2 + 0.023750 X_2^2 - 0.091250 X_3^2$$

The total effect of individual parameters on TA at linear, quadratic, and interactive levels is reported in Table 4 The F value (37.17) for the model was significant at 1%, i.e.,  $p < 0.01$  level of significance. Correlation coefficient  $R^2$  measures the generosity of fit of the regression model.  $R^2$  value for TA was 97.95%, which means that the model could account for 97.95% of data, and 2.05% variation is not elucidated by the model. The Pred R-Squared of 80.74 % was an in equitable agreement with the Adj R-Squared of 95.32 %. LOF was insignificant; hence, a second-order model was adequate in describing TA. The variation in TA value for debittered juice was observed from 0.84 to 0.99 (table 3). The TA score was found maximum (0.99) at the level of  $X_1$ (100: 0),  $X_2$  (50°C), and  $X_3$ (3 hrs).



**Figure 3.** 3D Response surface showing interaction effect of variables on TA of debittered *Citrus maxima* juice i) temperature and enzyme ratio ii) time and enzyme ratio and iii) time and temperature.

The 3D response surface curve for TA was presented in the figure. 3 (i, ii & iii). The values presented in Table 3 show that there were significant changes in TA of the debittered juice (0.84- 0.99). Figure 3 (i) shows that as temperature increase (30°C to 50°C) TA was first decreased slightly and then increased slightly in the presence of enzyme ratio. Figure 3 (ii) showed that the TA increased as time increase and maximum at a central point ( $X_3=3$  hr) and then decreased very rapidly as the time increased (3-4 hrs.) with different levels of enzyme ratio. From figure 3 (iii), it is clearly depicted that the TA increased as time increase and maximum at the central point ( $X_3=3$  hr) and then decreased very rapidly as the time increased (3-4 hrs.) with an increasing level of temperature. Our finding favors the finding of [44], who observed no significant TA changes during enzymatic treatment of pomegranate juice.

From table 6 of coefficient, it was observed that  $X_1$  (enzyme ratio) at 5% level, i.e.,  $p<0.05$  of significance,  $X_2$  (Temperature) at 1% level, i.e.,  $p<0.01$  of significance had a negative effect on TA at a linear level while  $X_3$  (Time) had a positive effect on TA at a linear level. Interactive effect of  $X_1$  (enzyme ratio),  $X_2$  (Temperature) had a significant negative effect at 5% level i.e.,  $p<0.05$  of significance and  $X_1$  (enzyme ratio),  $X_3$  (Time) and  $X_2$  (Temperature),  $X_3$  (Time) also had a significant negative effect at more than 10% level ( $p<0.1$ ) of significance. Quadratic effect of  $X_1$  (enzyme ratio) and  $X_3$  (Time) was negative at 5% level ( $p<0.05$ ) and 1% level ( $p<0.01$ ) of significance respectively, while  $X_2$  (Temperature) and was positive at 1 level % ( $p<0.01$ ) of significance.

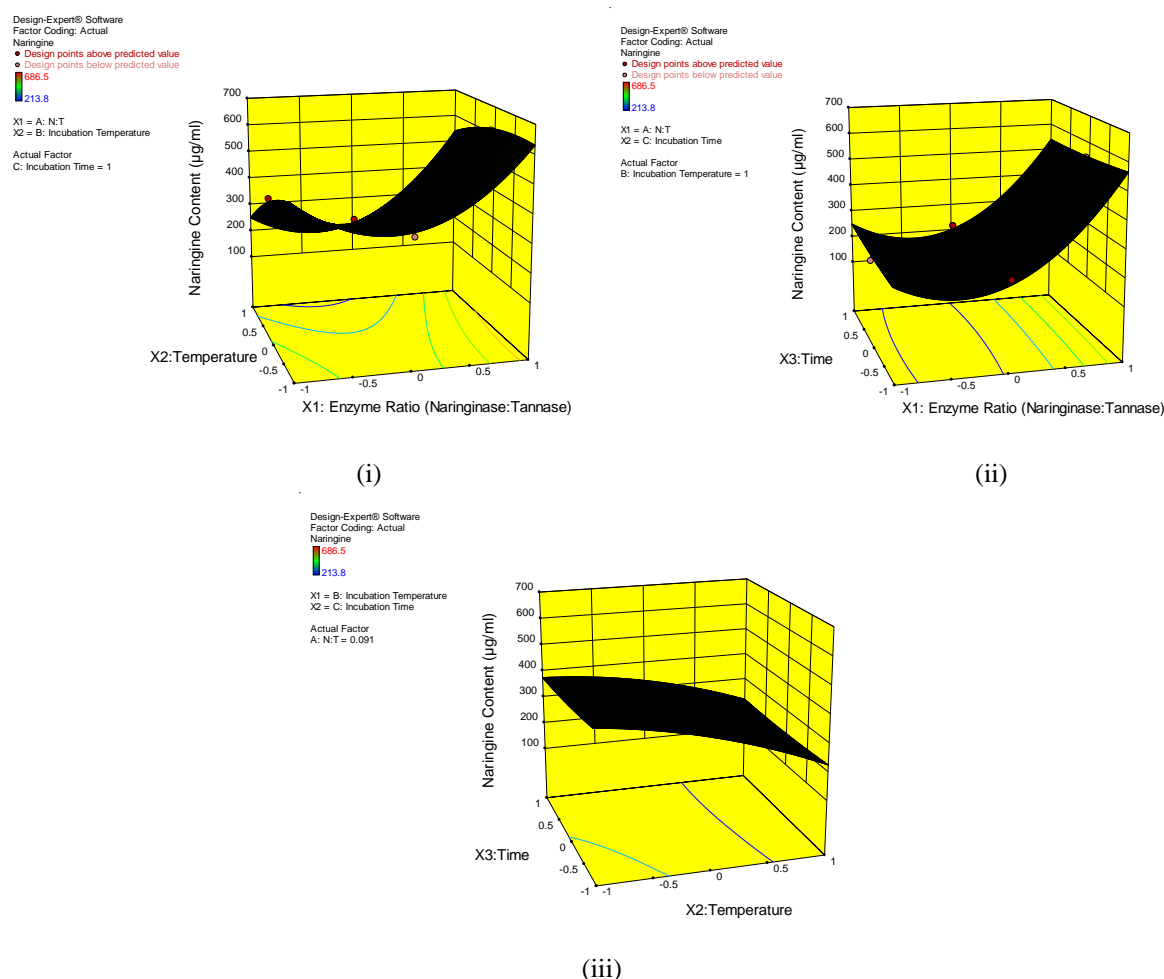
### *3.4 Effect on naringin content.*

Full second order equation to show the effect of  $X_1$ ,  $X_2$ , and  $X_3$  on naringin content could be explained by the equation given below.

$$\text{Naringin Content} = 328.80 + 118.7 X_1 - 95.08 X_2 - 15.44 X_3 + 37.45 X_1 X_2 - 10.52 X_1 X_3 + 12.90 X_2 X_3 + 190.99 X_1^2 - 32.19 X_2^2 + 6.69 X_3^2$$

The F value (42.54) for the model was significant at 1%, i.e.,  $p<0.01$  level of significance. Correlation coefficient R measures the generosity of fit of the regression model. The  $R^2$  value for naringin content was 98.20%, which means that the model could account for 98.20% of data, and 1.8% variation is not explained by the model. The Pred R-Squared of 88.48 % was in equitable agreement with the Adj R-Squared of 95.90 %. Lack of fit was insignificant; therefore, a second-order model was acceptable in describing naringin content. The variation in naringin content value for debittered juice was observed from 686.5 to 213.8 (table 3). The naringin content was found minimum (213.8) at the level of  $X_1$ (100: 0),  $X_2$  (50°C), and  $X_3$ (3 hrs).

The 3D response surface curve for naringin content was presented in figure 4 (i, ii & iii). The values presented in Table 3 show that there were significant changes in the naringin content of the debittered juice (686.5-213.8). Figure 4 (i) shows that as temperature increase (30°C to 50°C), naringin content was decreased significantly. With respect to enzyme ratio, naringin content was lower with naringinase enzyme. There was no decrement in naringin content with tannase enzyme in enzyme ratio. Figure 4 (ii) showed that the naringin content value was lower with the naringinase enzyme with time with respect to enzyme ration. Figure 4 (iii) clearly depicted that the naringin content decreased as time and temperature increased. A similar finding was observed by [46-49], who observed that the naringin content in grapefruit juice and pomelo juice respectively was decreased when the juice sample was treated with naringinase.



**Figure 4.** 3D Response surface showing interaction effect of variables on naringin content of debittered *Citrus maxima* juice i) temperature and enzyme ratio ii) time and enzyme ratio and iii) time and temperature.

From table 6 of coefficient, it was observed that  $X_1$  (enzyme ratio) had a positive effect on naringin content at 1% level, i.e.,  $p < 0.01$  of significance at a linear level, while  $X_2$  (Temperature) and  $X_3$  (Time) had a negative effect at 1% level, i.e.,  $p < 0.01$  and more than 10 % level, i.e.,  $p < 0.1$  of significance respectively. Interactive effect of  $X_1$  (enzyme ratio),  $X_2$  (Temperature), and  $X_3$  (Time) had a significant positive effect at 5% level ( $p < 0.05$ ) and more than 10 % level, i.e.,  $p < 0.1$  of significance respectively. Quadratic effect of  $X_1$  (enzyme ratio) and  $X_3$  (Time) was positive at 1% level ( $p < 0.01$ ) and more than 10 % level ( $p < 0.1$ ) of significance respectively, while  $X_2$  (Temperature) was negative at 10 % level ( $p < 0.1$ ) of significance.

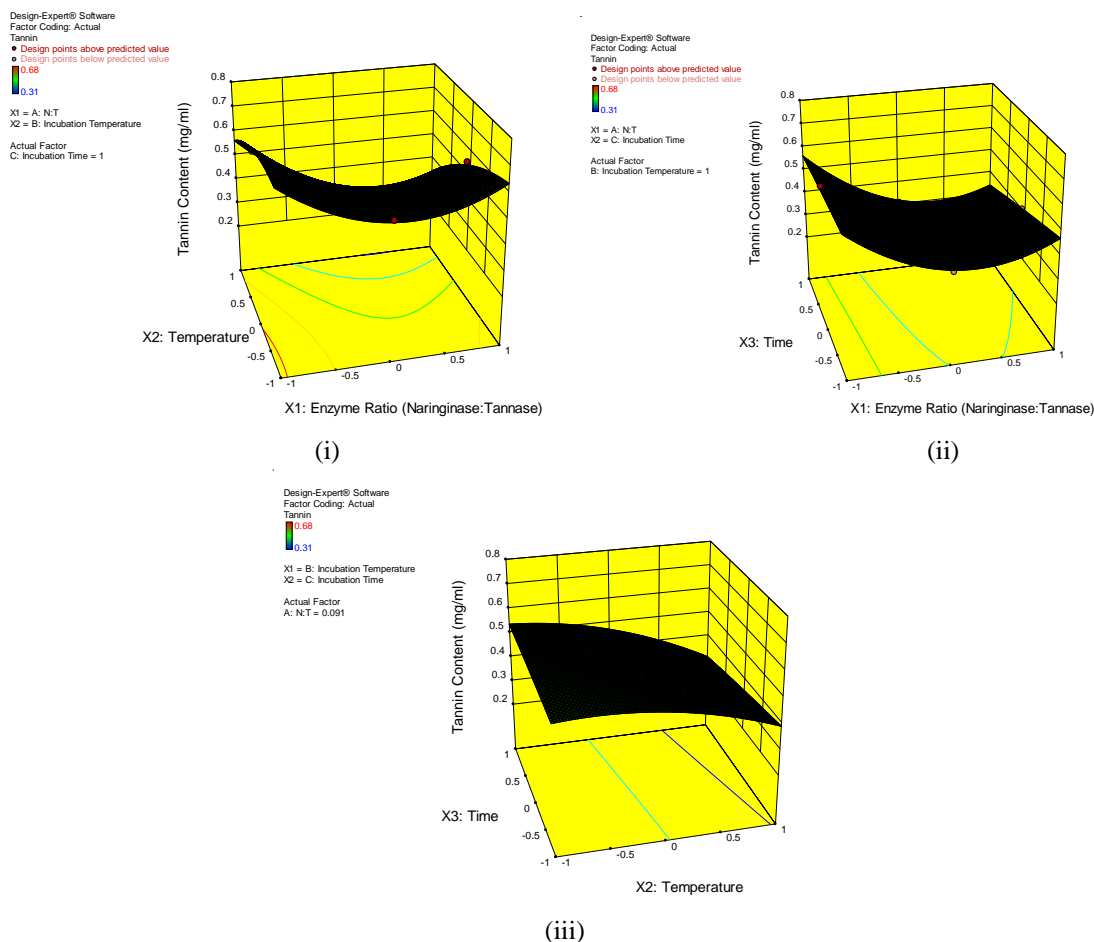
### 3.5 Effect on tannin content.

Complete second-order equation produced in terms of  $X_1$ ,  $X_2$ , and  $X_3$  for the effect of enzyme ratio, temperature, and time on tannin content could be explained by the equation given below.

$$\text{Tannin Content} = 0.49 - 0.050 X_1 - 0.084 X_2 - 0.011 X_3 - 0.035 X_1 X_2 - 0.025 X_1 X_3 - 0.028 X_2 X_3 + 0.13 X_1^2 - 0.048 X_2^2 - 3.250 \times 10^{-3} X_3^2$$

The F value (76.04) for the model was significant at a 1% ( $p < 0.01$ ) level of significance. Correlation coefficient  $R^2$  measures the generosity of fit of the regression model. The  $R^2$  value for tannin content was 98.90 %, which means that the model could account for 98.90% of data, and the model does not elucidate 1.1% variation. The  $R^2$  value higher than 90% meant that the

regression model explained the reaction well. The Pred R-Squared of 86.48 % was in equitable agreement with the Adj R-Squared of 97.69 %. Lack of fit was insignificant; hence, a second-order model was acceptable in describing naringin content. The tannin content value variation for debittered juice was observed from 0.68 to 0.31 (table 3). The tannin content was found minimum (0.31) at the level of  $X_1(50: 50)$ ,  $X_2(50^\circ\text{C})$ , and  $X_3(4\text{ hrs})$ .



**Figure 5.** 3D Response surface showing interaction effect of variables on the tannin content of debittered *Citrus maxima* juice i) temperature and enzyme ratio ii) time and enzyme ratio and iii) time and temperature.

The 3D response surface curve for tannin content was presented in figure 5 (i, ii & iii). The values presented in Table 3 show that there were significant changes in the tannin content of the debittered juice (0.68- 0.31). Figure 5 (i) shows that as temperature increase ( $30^\circ\text{C}$  to  $50^\circ\text{C}$ ), tannin content was decreased significantly. With respect to enzyme ratio, there was no change in tannin content with the naringinase enzyme. There was a significant change in tannin content with tannase enzyme in enzyme ratio. Figure 5 (ii) showed that the tannin content decreased significantly with an increase in time, and tannin content was also decreased with tannase enzyme in enzyme ratio. Figure 5 (iii), it is depicted that the tannin content decreased as temperature increased. Our findings favor the findings of [50-52].

From table 6 of coefficient, it was observed that  $X_1$  (enzyme ratio),  $X_2$  (Temperature), and  $X_3$  (Time) had a negative effect on tannin content at 1% level ( $p < 0.01$ ), 1% level ( $p < 0.01$ ) and 10 % level ( $p < 0.1$ ) of significance respectively. Interactive effect of  $X_1$  (enzyme ratio),  $X_2$ (Temperature) and  $X_1$  (enzyme ratio),  $X_3$  (Time) and  $X_2$  (Temperature),  $X_3$  (Time) had significant negative effect at 1% level ( $p < 0.01$ ), 5% level ( $p < 0.05$ ) and 5 % level ( $p < 0.05$ ) of significance respectively. Quadratic effect of  $X_1$  (enzyme ratio) was positive at 1% level

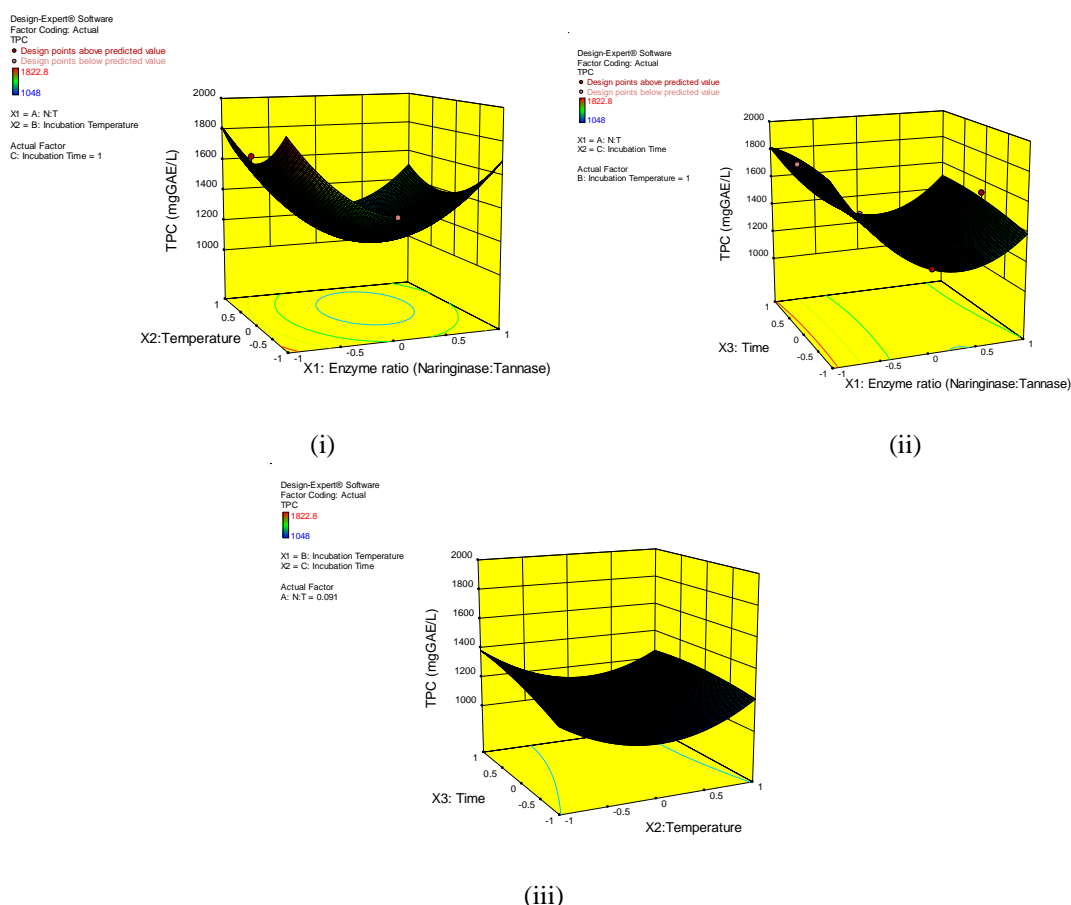
( $p < 0.01$ ) of significance. In comparison,  $X_2$  (Temperature) was negative at 1% level ( $p < 0.01$ ) of significance, but the quadratic effect of  $X_3$  (Time) was positive at more than 10 % level ( $p < 0.1$ ) of significance.

### 3.6. Effect on TPC.

Full second order equation to show the effect of  $X_1$ ,  $X_2$ , and  $X_3$  on TPC could be explained by the equation given below.

$$\text{TPC} = 1109.64 - 163.41 X_1 - 27.90 X_2 + 53.34 X_3 - 29.25 X_1 X_2 + 37.57 X_1 X_3 - 33.90 X_2 X_3 + 393.49 X_1^2 + 195.42 X_2^2 - 29.01 X_3^2$$

The F value (47.76) for the model was significant at 1%, i.e.,  $p < 0.01$  level of significance. Correlation coefficient R measures the goodness of fit of the regression model.  $R^2$  value for TPC was 98.40%, which implies that the model could account for 98.40% of data, and the model does not explain 1.6% variation. The Pred R-Squared of 80.47 % was in equitable agreement with the Adj R-Squared of 96.34 %. Lack of fit was insignificant; therefore, a second-order model was acceptable in describing naringin content. The variation in TPC value for debittered juice was observed from 1048 to 1822.8 (table 3). The TPC was found minimum (1048.0) at the level of  $X_1$ (50: 50),  $X_2$  (40°C), and  $X_3$ (3 hrs).



**Figure 6.** 3D Response surface showing interaction effect of variables on TPC of debittered *Citrus maxima* juice, i) temperature and enzyme ratio ii) time and enzyme ratio and iii) time and temperature.

The 3D response surface curve for TPC was presented in figure 6 (i, ii & iii). The values presented in Table 3 show that there were significant changes in TPC of the debittered juice (1048.0 – 1822.8). Figure 6 (i) shows that TPC increased with the enzyme ratio as temperature increase (30°C to 50°C). Figure 6 (ii) showed that the TPC increased significantly with an

increase in time while decreased with tannase enzyme with respect to enzyme ratio. Figure 6 (iii) depicted that the TPC increased as incubation time and incubation temperature increased. This increment in TPC could be co-related with the presence of naringenin (i.e., a metabolite produced from naringin with hydroxyl groups) or, due to the possibility of overestimation of polyphenolic content with respect to lack of selectivity of Folin-Ciocalteu's reagent [53], when analyzed through the spectrophotometric method. It also shows that Folin-Ciocalteu's reagent reacts with reducing compounds like sugars (e.g., glucose) and not only reacts with phenols[54]. Our findings favor the findings of Cavia-Saiz [55, 49], who observed that the total phenolic content was increased as the naringinase concentration increases.

From table 6 of coefficient, it was observed that  $X_1$  (enzyme ratio) and  $X_2$  (Temperature) had a negative effect on TPC at 1% level ( $p < 0.01$ ) and more than 10 % level ( $p < 0.1$ ) of significance, respectively while  $X_3$  (Time) had positively effect on TPC at 5% level ( $p < 0.05$ ) of significance. Interactive effect of  $X_1$  (enzyme ratio),  $X_2$  (Temperature) and  $X_3$  (Incubation Time) had a significant negative effect at more than 10% level ( $p < 0.1$ ) of significance while  $X_1$  (enzyme ratio),  $X_3$  (Incubation Time) was positive at more than 10% level ( $p < 0.1$ ) of significance. Quadratic effect of  $X_1$  (enzyme ratio) and  $X_2$  (Temperature) was positive at 1% level ( $p < 0.01$ ) of significance, while  $X_3$  (Time) was negative at more than 10 % level ( $p < 0.1$ ) of significance.

### 3.7. Effect on vitamin C content.

Complete second-order equation produced in terms of  $X_1$ ,  $X_2$ , and  $X_3$  for the effect of enzyme ratio, temperature, and time on Vitamin C content could be explained by the equation given below.

$$\text{Vitamin C Content} = 25.54 - 2.11 X_1 - 3.76 X_2 - 2.58 X_3 - 0.4 X_1 X_2 - 2.17 X_1 X_3 - 0.63 X_2 X_3 + 20.91 X_1^2 + 6.85 X_2^2 + 5.13 X_3^2$$

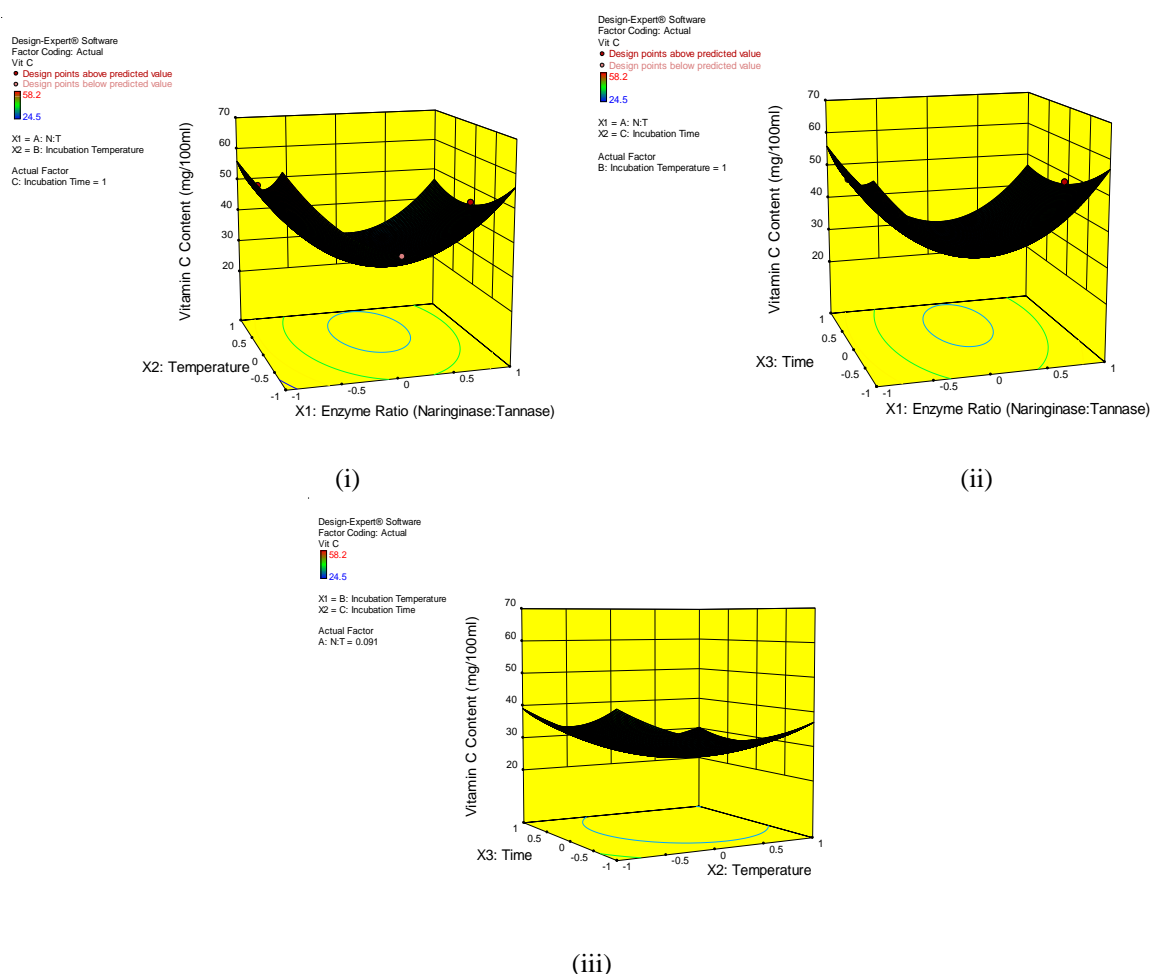
The F value (147.9) for the model was significant at 1% ( $p < 0.01$ ) level of significance. Correlation coefficient  $R^2$  measures the generosity of fit of the model. The  $R^2$  value for TPC was 99.48%, which means that the model could account for 99.48% of data, and the model does not elucidate 0.52% variation. The Pred R-Squared of 94.68 % was in equitable agreement with the Adj R-Squared of 98.80 %. Lack of fit was insignificant; hence, a second-order model was acceptable in describing Vitamin C content. The variation in vitamin C content value for debittered juice was observed from 58.2 to 24.5 (table 3). The vitamin C content was found maximum (58.2) at the level of  $X_1$ (100: 0),  $X_2$  (30°C), and  $X_3$ (3 hrs).

The 3D response surface curve for vitamin C content was presented in figure 7 (i, ii & iii). The values presented in table 4 show that there were significant changes in vitamin C content of the debittered juice (24.5 - 58.2). Figure 7 (i) shows that as temperature increase (30°C to 50°C), vitamin C content was decreased with enzyme ratio. Figure 7 (ii) showed that vitamin C content decreased significantly with an increase in enzyme ratio time. Figure 7 (iii) depicted that the vitamin C content decreased as time and temperature increased. The vitamin C content was found to increase near the central value of enzyme concentration. The decrease in response (ascorbic acid content) may be due to the denaturation of the enzyme at high temperatures. A similar finding was reported by [56-57], who observed that the ascorbic acid content decreases as the temperature and time increases, respectively.

From table 6 of the coefficient, it was observed that  $X_1$  (enzyme ratio),  $X_2$  (Temperature), and  $X_3$  (Time) had a negative effect on vitamin C content at a 1% level ( $p < 0.01$ ) of significance. The interactive effect of  $X_1$  (enzyme ratio),  $X_3$  (Time) had a significant



negative effect at a 5% level ( $p < 0.05$ ) of significance. Quadratic effect of  $X_1$  (enzyme ratio),  $X_2$  (Temperature), and  $X_3$  (Time) were positive at 1% level ( $p < 0.01$ ) of significance.



**Figure 7.** 3D Response surface showing interaction effect of variables on Vitamin C content of debittered *Citrus maxima* juice i) temperature and enzyme ratio ii) time and enzyme ratio and iii) time and temperature.

## 4. Conclusions

The model F value was found to be highly significant at a 1% level of significance for all the responses. The second-order model could be fitted to predict the entire dependent parameters. The optimum values for debittering and clarification of citrus maxima juice were found to be 54.55:45.45, 50°C, and 4 hrs. The values for pH, TSS, TA, Naringin content, Tannin content, TPC and Vitamin C content were found to be 3.17, 6.256 °Brix, 0.885 % citric acid, 220.549 µg/ml, 0.311 mg/ml, 1256.721 mg GAE/L, 30.309 mg/100ml respectively. It could be concluded that the enzymatic debittering and clarification of citrus maxima juice could be done by immobilized enzymes naringinase and tannase, which is produced from *Aspergillus* sp. SK01 isolated from rotten Citrus maxima under optimum processing conditions.

## Funding

This research received no external funding.



## Acknowledgments

The authors are thankful to the Department of Life Sciences, Graphic Era (Deemed-to-be) University, Dehradun, Uttarakhand, India, for permitting us to carry out the study at the University.

## Conflicts of Interest

The authors declare no conflict of interest.

## References

1. Morton, J. Pummelo *Citrus maxima*: Fruits of warm climates. Pummelo, **1987**; pp.147–151.
2. Burda, S.; Oleszek, W. Antioxidant and antiradical activities of flavonoids. *J Agri Food Chem* **2001**, *49*, 2774–2779, <https://doi.org/10.1021/jf001413m>.
3. Fattouch, S.; Caboni, P.; Coroneo, V. Antimicrobial activity of Tunisian quince (*Cydonia oblonga* Miller) pulp and peel polyphenolic extracts. *Journal of Agriculture and Food Chemistry* **2007**, *55*, 963–969, <https://doi.org/10.1021/jf062614e>.
4. Montanari, A.; Chen, J.; Widmer, W. Citrus flavonoids: A review of past biological activity against disease. In: *Flavonoids in the Living System*. Manthey, J.A.; Buslig, B.S. (Eds.), New York: Plenum Press, **1998**; pp. 103–113, <https://doi.org/10.1007/978-1-4615-5335-9>
5. Samman, S.; Wall, P. M. L.; Cook, N. C. Flavonoids and coronary heart disease: Dietary perspectives. In: *Flavonoids in the Living System*. Manthey, J.A.; Buslig, B.S. (Eds.), New York: Plenum Press, **1996**; pp. 469–481.
6. Peace, N.A.; Happiness, C.A. Nutrient, phytochemical, and anti-nutrient composition of Citrus maxima fruit juice and peel extract. *Food Science and nutrition* **2018**, *6*, 653–658, <https://doi.org/10.1002/fsn3.604>.
7. Li, L.; Tan, W.; Li, W.; Zhu, Y.; Cheng, Y.; Ni, H. Citrus taste modification potentials by genetic engineering. *Int. J. Mol. Sci.* **2019**, *20*, <https://doi.org/10.3390/ijms20246194>.
8. Gupta, A.K.; Koch, P.; Mishra, P. Optimization of debittering and deacidification parameters for Pomelo juice and assessment of juice quality. *J Food Sci Technol* **2020**, *57*, 4726–4732, <https://doi.org/10.1007/s13197-020-04687-w>.
9. Awad, G.E.A.; Abd El Aty, A.A.; Shehata, A.N.; Hassan, M.E.; Elnashar, M.M. Covalent immobilization of microbial naringinase using novel thermally stable biopolymer for hydrolysis of naringin. *3 Biotech* **2016**, *6*, <https://doi.org/10.1007/s13205-015-0338-x>
10. Hasegawa, S.; Maier, V.P. Solution to the limonin bitterness problem of citrus juices. *Food Technology* **1993**, *37*, 73–77.
11. Pereira, G.A.; Arruda, H.S.; Morais, D.R.; Araujo, N.M.P.; Pastore, G.M. Mutamba (*Guazuma ulmifolia* Lam.) fruit as a novel source of dietary fibre and phenolic compounds. *Food Chem* **2020**, *310*, <https://doi.org/10.1016/j.foodchem.2019.125857>.
12. Singh, S.S.; Abdullah, S.; Pradhan, R.C.; Mishra, S. Physical, chemical, textural, and thermal properties of cashew apple fruit. *Journal of Food Process Engineering* **2019**, *42*, 1–10, <https://doi.org/10.1111/jfpe.13094>.
13. Singh, J.; Kundu, D.; Das, M.; Banerjee, R. Enzymatic processing of juice from fruits/vegetables: An emerging trend and cutting edge research in food biotechnology. In: *Enzymes in food biotechnology*. San Diego, California, USA: Academic Press **2019**; pp. 419–432, <https://doi.org/10.1016/B978-0-12-813280-7.00024-4>.
14. Aguilar, C.; Rodriguez, R.; Gutierrez-Sanchez, G.; Augur, C.; Favela-Torres, E.; Prado-Barragan, L.A.; Ramirez-Coronel, A.; Coterias-Esquivel, J.C. Microbial Tannases: advances and perspectives. *Appl. Microbiol. Biotechnol* **2007**, *76*, 47–59, <https://doi.org/10.1007/s00253-007-1000-2>
15. Rout, S.; Banerjee, R. Production of tannase under mSSF and its application in fruit juice debittering. *Indian J Biotechnol* **2006**, *5*, 351–356.
16. Patil, S.V.; Koli, S.H.; Mohite, B.V.; Patil, R.P.; Patil, R.R.; Borase, H.P.; Patil, V.S. A novel screening method for potential naringinase producing microorganisms. *Biotechnol. Appl. Biochem* **2019**, *66*, 323–327, <https://doi.org/10.1002/bab.1728>.
17. Arshad, R.; Mohyuddin, A.; Saeed, S.; Hassan, A.U. Optimized production of tannase and gallic acid from fruit seeds by solid state fermentation. *Tropical Journal of Pharmaceutical Research* **2019**, *18*, 911–918.
18. Chandler, B.V.; Nicol, K.J. *Some relationships of naringin: their importance in orange juice bitterness*. CSIRO Food Res Quart Volume 35, **1975**; pp.79–88.
19. Habelt, K.; Pittner, F. A rapid method for the determination of naringin, prunin, and naringin applied to the assay of naringinase. *Anal Biochem* **1983**, *134*, 393–397, [https://doi.org/10.1016/0003-2697\(83\)90314-7](https://doi.org/10.1016/0003-2697(83)90314-7).

20. Purewal, S.S.; Sandhu, K.S. Debittering of citrus juice by different processing methods: A novel approach for food industry and agro-industrial sector *Scientia Horticulturae* **2021**, *276*, <https://doi.org/10.1016/j.scienta.2020.109750>.
21. Yadav, V.; Yadav, P.K.; Yadav, S.; Yadav, K.D.S.  $\alpha$ -l- Rhamnosidase: A review. *Process Biochem* **2010**, *45*, 1226–1235, <https://doi.org/10.1016/j.procbio.2010.05.025>.
22. Ono, M.; Tgsa, T.; Chibata, I. Preparation and properties of immobilized naringinase using tannin-aminohexyl cellulose. *Agric. Biol. Chem* **1978**, *42*, 1847–1853, <https://doi.org/10.1080/00021369.1978.10863264>.
23. Vila-Reala, H.; Alfaia, A.J.; Rosa, M.E.; Calado, A.R.; Ribeiro, M.H.L. An innovative sol-gel naringinase bioencapsulation process for glycosides hydrolysis. *Process Biochem* **2010**, *45*, 841–850, <https://doi.org/10.1016/j.procbio.2010.02.004>.
24. Ribeiro, M.H. Naringinase: occurrence, characteristics, and applications. *Appl. Microbiol. Biotechnol* **2011**, *90*, 1883–1895. <https://doi.org/10.1007/s00253-011-3176-8>.
25. Sharma, K.P. Tannin degradation by phytopathogen's tannase: A Plant's defense perspective. *Biocatalysis and Agricultural Biotechnology* **2019**, *21*, <https://doi.org/10.1016/j.bcab.2019.101342>.
26. Purohit, J.S.; Dutta, J.R.; Nanda, R.K.; Banerjee, R. Strain improvement for tannase production from co-culture of *Aspergillus foetidus* and *Rhizopus oryzae*. *Bioresource Technol* **2006**, *97*, 795-801, <https://doi.org/10.1016/j.biortech.2005.04.031>.
27. Abd El Tawab, A.M.; Murad, H.A.; Khattab, M.S.A.; Azzaz, H.H. Optimizing Production of Tannase and in vitro Evaluation on Ruminant Fermentation, Degradability and Gas Production. *International Journal of Dairy Science* **2019**, *14*, 53-60. <https://doi.org/10.3923/ijds.2019.53.60>.
28. Puri, M.; Banerjee, A.; Banerjee, U.C. Optimization of process parameters for the production of naringinase by *Aspergillus niger* MTCC 1344. *Process Biochem* **2005**, *40*, 195–201, <https://doi.org/10.1016/j.procbio.2003.12.009>.
29. Puri, M.; Kalra, S. Purification and characterization of naringinase from a newly isolated strain of *Aspergillus niger* 1344 for the transformation of flavonoids. *World Journal of Microbiology & Biotechnology* **2005**, *21*, 753–758, <https://doi.org/10.1007/s11274-004-5488-7>.
30. Pegu, B.K.; Chutia, J.; Kardong, D.; Gogoi, D. Optimization of environmental parameters for enhancement of naringinase production of *Bacillus cereus*-K1 a bacterial strain. *Int. J. Adv. Sci. Res. Manag* **2019**, *4*, 68–73.
31. Costa da A.M.; Kadowaki, M.K.; Minozzzo, M.C.; de souza, C.G.M.; Boer, C.G.; Brcht, A.; Marina, R. Production, purification and characterization of tannase from *Aspergillus tamarii*. *African Journal of Biotechnology* **2012**, *11*, 391-398.
32. Davis, W.B. Determination of flavanones in citrus industry. *Anal. Chem.* **1947**, *19*, 476-478, <https://doi.org/10.1021/ac60007a016>.
33. Mondal, K.C.; Banerjee, D.; Jana, M.; Pati, B.R. Colorimetric assay for determination of tannin acyl hydrolase (E.C. 3.1.1.20) activity. *Anal. Biochem* **2001**, *295*, 168–171, <https://doi.org/10.1006/abio.2001.5185>.
34. Elnashar, M.M. Review article: immobilized molecules using biomaterials and nanobiotechnology. *J Biomater Nanobiotechnol* **2010**, *1*, 61–76, <https://doi.org/10.4236/jbnt.2010.11008>.
35. Srivastava, A.; Kar, R. Application of immobilized tannase from *Aspergillus niger* for the removal of tannin from myrobalan juice. *Indian J. Microbiol* **2010**, *50*, 41-56, <https://doi.org/10.1007/s12088-010-0029-6>.
36. Rangana, S. *Analysis and quality control for fruit and vegetable products*. Tata McGraw Hill Education Pvt. Ltd., New Delhi. **2010**.
37. Sawhney, S. K. ; Singh, R. Estimation of ascorbic acid in lemon juice, Introductory practical Biochemistry. Narosa Publishing House, New Delhi. **2015**; pp.104-105.
38. Kohli, D.; Kumar, A.; Kumar, S.; Upadhyay, S. Waste Utilization of Amla Pomace and Germinated Finger Millets for Value Addition of Biscuits. *Current Research in Nutrition and Food Science* **2019**, *7*, 272-279, <https://doi.org/10.12944/CRNFSJ.7.1.27>.
39. Makkar, H.P.S.; Bluemmel, M.; Borowy, N.K.; Becker, K. Gravimetric determination of tannins and their correlations with chemical and protein precipitation methods. *Journal of Science Food Agriculture* **1993**, *61*, 161–165.
40. Hagerman, A.E.; Butler, L.G. Protein precipitation method for the quantitative determination of tannins. *J. Agric. Food Chem* **1978**, *26*, 809-812, <https://doi.org/10.1021/jf60218a027>.
41. Kumar, S.; Khadka, M.; Mishra, R.; Kohli, D.; Upadhaya, S. Effects of Conventional and Microwave Heating Pasteurization on Physiochemical Properties of Pomelo (*Citrus maxima*) Juice. *J Food Process Technol* **2017**, *8*.
42. Cornell, J.A. Experiments with mixtures: Designs, models, and the analysis of mixture data. 3rd ed., Wiley Series, In: *Probability And Statistics*. **2011**; <https://doi.org/10.1002/9781118204221>.
43. Suri, S.; Dutta, A.; Shahi, N.C.; Raghuvanshi, R.S.; Singh, A.; Chopra, C.S. Numerical optimization of process parameters of ready-to-eat (RTE) iron rich extruded snacks for anemic population. *LWT - Food Science and Technology* **2020**, *134*, <https://doi.org/10.1016/j.lwt.2020.110164>.

44. Massimiliano, R.; Augusta, C.; Rossana, B.; Gerardo, P.; Davide, B.; Roberto, M. The effect of fruit processing and enzymatic treatments on pomegranate juice composition, antioxidant activity and polyphenols content. *LWT - Food Science and Technology* **2013**, *53*, 355-359, <https://doi.org/10.1016/j.lwt.2013.02.015>.
45. Ashima, K.; Hina, I. Efficiency of Tannase Produced by *Trichoderma Harzianum* MTCC 10841 in Pomegranate Juice Clarification and Natural Tannin Degradation. *International Journal of Biotechnology and Bioengineering Research* **2006**, *4*, 641-650.
46. Prakash, S.; Singhal, R.S.; Kulkarni, P.R. Enzymatic debittering of Indian grapefruit (*Citrus paradise*) juice. *Journal of the Science of Food and Agriculture* **2002**, *82*, 394-397, <https://doi.org/10.1002/jsfa.1059>.
47. Ghosh, U.; Gangopadhyay, H. Enzymatic, physicochemical and rheological behaviour of bael fruit pulp and juice. *Indian Journal of Chemical Technology* **2002**, *9*, 123-126.
48. Patil, M.B.; Dhake, A.B. Debittering of citrus fruit juice by naringinase of *Penicillium purpurogenum*. *Int. J. Engg. Res. & Sci. & Tech* **2014**, *3*, 266-270.
49. Singla, G.; Panesar, P.S.; Sangwan, R.S.; Krishania, M. Enzymatic processing of *Citrus reticulata* (Kinnow) pomace using naringinase and its valorization through preparation of nutritionally enriched pasta. *J Food Sci Technol* **2020**, <https://doi.org/10.1007/s13197-020-04846-z>.
50. Lima, J.S.; Cruz, R.; Fonseca, J.C.; Medeiros, E.V.; Maciel, M.H.C.; Moreira, K.A.; Motta, C.M.S. Production, characterization of tannase from *Penicillium montanense* URM 6286 under ssf using agroindustrial wastes, and application in the clarification of grape juice (*Vitis vinifera* L.). *Sci. World J* **2014**, *2014*, 1-9, <http://dx.doi.org/10.1155/2014/182025>.
51. da Silva, V.M.A.; Cruz, R.; Fonseca, J.C.; de Souza-Motta, C.M.; de Sena, A.R.; Moreira, K.A. Juice clarification with tannases from *Aspergillus carneus* URM5577 produced by solid-state fermentation using *Terminalia catappa* L. leaves. *African Journal of Biotechnology* **2017**, *16*, 1131-1141, <https://doi.org/10.5897/AJB2017.15958>.
52. Abdullah, S.; Pradhan, R.C.; Aflah, M.; Mishra, S. Efficiency of tannase enzyme for degradation of tannin from cashew apple juice: Modeling and optimization of process using artificial neural network and response surface methodology. *J Food Process Eng* **2020**, *10*, <https://doi.org/10.1111/jfpe.13499>.
53. Escarpa, A.; González, M.C. Approach to the content of total extractable phenolic compounds from different food samples by comparison of chromatographic and spectrophotometric methods. *Analytica Chimica Acta* **2001**, *427*, 119–127, [https://doi.org/10.1016/S0003-2670\(00\)01188-0](https://doi.org/10.1016/S0003-2670(00)01188-0).
54. Singleton, V.L.; Rossi, J.A. Colorimetric of total phenolics with phosphomolybdic-phosphotungstic acid reagent. *American Journal of Enology and Viticulture* **1965**, *16*, 144–158.
55. Cavia-Saiz, M.; Muñoz, P.; Ortega, N.; Busto, M.D. Effect of enzymatic debittering on antioxidant capacity and protective role against oxidative stress of grapefruit juice in comparison with adsorption on exchange resin. *Journal Food Chemistry* **2011**, *125*, 158–163, <https://doi.org/10.1016/j.foodchem.2010.08.054>.
56. Sawinder, K.; Sarkar, B.C.; Sharma, H.K.; Charanjiv, S. Response Surface Optimization of Conditions for the Clarification of Guava fruit Juice Using Commercial Enzyme. *Journal of Food Process Engineering* **2011**, *34*, 1298–1318, <https://doi.org/10.1111/j.1745-4530.2009.00414.x>.
57. Essodolom, P.; Chantal, B.E.; Mamatchi, M.; Kous'anta, A. Effect of temperature on the degradation of ascorbic acid (vitamin c) contained in infant supplement flours during the preparation of porridges. *International Journal of Advanced research* **2020**, *8*, 116-121, <http://dx.doi.org/10.21474/IJAR01/10605>.