


Rheological Behavior Models of Polymers

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Abstract: We studied and investigated the various viscosimetric and rheological polymers' behaviors during this comprehensive review. The viscosities relate to the investigation of the flux, the deformation, and the polymers' elasticity; we have employed the viscosity since this plays a primordial role in the phenomena flux and implementation of the polymer. The rheology behaviors were investigated for the determination of the physical properties of polymers. The rheological properties are mostly employed for improving polymers implementation. Further, three rheological behaviors models such as Newtonian, pseudo-plastic (Power Law, Law of Tile and Cross Law) and heat-dependent pseudo-plastic (Williams-Landel-Ferry Law (WLF), Law of Tile-Yasuda and Arrhenius law) were studied.

Keywords: comprehensive review; rheological behaviors; polymers; Newtonian and pseudo-plastic models.

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1. Introduction

Polymer composite materials (a mixture of two or more polymers) are an excellent alternative to the synthesis of polymers with improved properties and may exhibit combinations of properties in general superior to those of pure components [1-3]. Then, polymer composites can be elaborated by conventional machines used in industry, such as extruders, internal mixers, and injection molding machines [4-8]. Composite polymers can be used as low shear rate Newtonian fluids, generally in the vicinity of 1 S^{-1} , and as high shear rate non-Newtonian fluids. In the injection molding process, the shear rate generally reaches 106 S^{-1} , which is well beyond the traditional rheometer limits [9-14]. Many viscometers have been developed according to the fluid's nature to be studied [15-19]. Devices employed for extruders are prepared with thin matrices that allow elasticities and viscosities to be determined but at a low shear rate. The purpose of the original rheometer on injection or extrusion polymer processing machines is to allow the determination of various rheological properties such as viscosity, elasticity, and/or compressibility, under actual processing conditions, without disturbing the processing cycle, in a minimum of time and with a high degree of accuracy [20-25]. It can be used both as a device for measuring and determining the rheological behavior laws of plastics or as a machine control and regulation device, in which case it can be controlled directly by the machine's microprocessor [26-31]. Rheology is the technology that studies the distortion of bodies under the effect of speed stresses. Also, it studies the relationship between stress and strain as a function of time in the material [32-37]. Generally speaking, rheology

assumes that material is continuous (no voids in the material), each point of the body moves continuously, two points of the body that are infinitely adjacent before the deformation are still infinitely adjacent after the deformation and physical properties of the body vary continuously from one point to another, respectively. Rheology can be classified into two types [38-43]: Experimental rheology determines the behavioral relationship between stress and strain rate experimentally; Theoretical rheology provides a limited number of mathematical models of the behavior independently of the microscopic structure;

Gaseous, liquid, or solid bodies are divided into two classes as Newtonian fluids (these are all gases and a large number of liquids; their rheological equation is simple these bodies have only one rheological characteristic: viscosity) and bodies with complicated rheological equations: these are non-Newtonian liquids and solids [44-48].

Therefore, in the case of polymers rheology, only studies displacements that are large relative to the size of the macromolecules. Intramolecular movements and chain entanglement are movements that rheology by nature cannot describe. Then, these local movements are the basis for explaining the rheological behavior of molten polymers [49-53].

Viscosity is a very important parameter in the study of polymer flow. The latter, which is about 10^6 to 10^8 times that of water, is a function of certain parameters such as weight of the polymer, temperature, pressure and shear rate, respectively.

2. Rheological Behavior in Dynamic Regime (Low Deformations)

defined The strain $\gamma(t)$ and the stress $\sigma(t)$ were defined according to the following relations (Equations 1 and 2) [28, 29]:

$$\gamma(t) = \gamma_0 e^{i\omega t} \quad (1)$$

$$\tau(t) = \tau_0 e^{i(\omega t + \delta)} \quad (2)$$

Where γ_0 and τ_0 denote the maximum amplitudes of the strain and stress, respectively, t is time. Further, the complex modulus G^* was calculated from strain and stress according to Equation 3 [28, 29]. Also, the complex modulus G^* was composed of two parts, such as a real part (G') and an imaginary part (G'') according to the following Equation 4. Then, The loss factor and the complex viscosity η^* were measured according to the following Equations 5 and 6 [28, 29]:

$$G^* = \frac{\tau(t)}{\gamma(t)} = \frac{\tau_0}{\gamma_0} e^{i\delta} \quad (3)$$

$$G^* = G' + iG'' \quad (4)$$

$$\tan\delta = \frac{G''}{G'} \quad (5)$$

$$\eta^* = \eta' + i\eta'' \quad (6)$$

2.1. Rheological behavior in the statistical regime.

In statistical regime, the viscosity η was determined from shear stress τ and speed deformation $\dot{\gamma}$ (Equation 7) [28, 29].

$$\eta = \frac{\tau}{\dot{\gamma}} \quad (7)$$

The viscosity don't depend to the speed of deformation, indicating for a Newtonian fluid, the stress is studied according to the deformation's speed. We obtained a line whose value of the slope is the viscosity, noted η_0 (Figure 1-a). When the viscosity evolves according to the speed of deformation, the fluid's behavior is called pseudo-plastic (Figure 1-b). In these two cases, these fluids deform immediately shear stress is applied. If the material does not deform (or little) below a certain, this is called threshold stress. The refore, the threshold constraint can be defined by the following relation (Equation 8) [28, 29].

$$\tau_y = \lim_{\dot{\gamma} \rightarrow 0} \sigma(\dot{\gamma}) \quad (8)$$

Viscoelastic behaviors present threshold stress. At the high stress compared to the threshold stress, the viscosity does not depend on the viscosity of strain (Figure 1-c) [28, 29].

2.2. Rheological behaviors.

Thermosetting polymers are not Newtonian and their apparent viscosity decreasing according to the shear rate. This behavior is the pseudo-plasticity type at lower temperatures, the viscosity of polymer increases (the polymer is viscous). The flow speed of the polymer thermosetting between two edges changes, this variation was owing to the viscous polymers. Viscosity variation of polymers as a function of the shear rate and temperature are shown in Figure 2 [31].

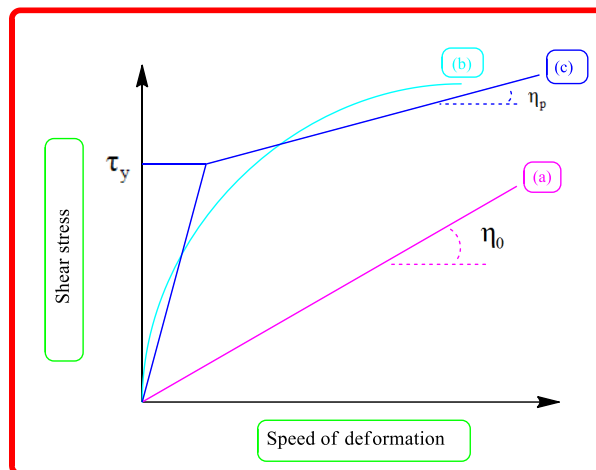


Figure 1. Varying rheological models: (a) Newtonian, (b) pseudo-plastic and (c) viscoelastic.

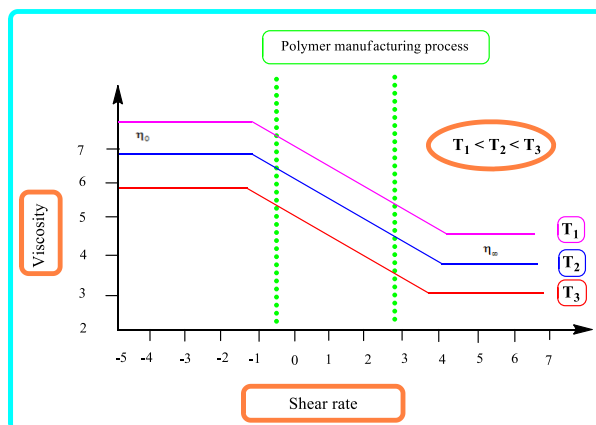


Figure 2. Viscosity as a function of the shear rate at different temperatures.

2.3. Varying viscosities models.

Viscosities models most employed are Newtonian model, pseudo-plastic model and heat-dependent pseudo-plastic model, respectively.

2.3.1. Newtonian model.

The relation between stress and sheer speed presents the viscosity. Stress speed u is proportional to this speed $F = Ku$. The components of the velocity vector are shown in the following equation 9 [28, 29, 31]:

$$\vec{u} \begin{cases} u = \dot{\gamma} \cdot y \\ v = 0 \\ w = 0 \end{cases} \quad (9)$$

This force related to the surface on which it is exerted the shear stress $\tau = F/S$ proportional to the sheer speed (Equation 10):

$$\tau = \eta \cdot \dot{\gamma} \quad (10)$$

The proportionality coefficient between τ and $\dot{\gamma}$ was defined as a dynamic viscosity η (Equation 11).

$$\eta = \frac{\tau}{\dot{\gamma}} \quad (11)$$

The shear stress as a function of the speed gradient is shown in Figure 3-a. However, The viscosity versus speed gradient shown a straight line parallel to the X-axis [31].

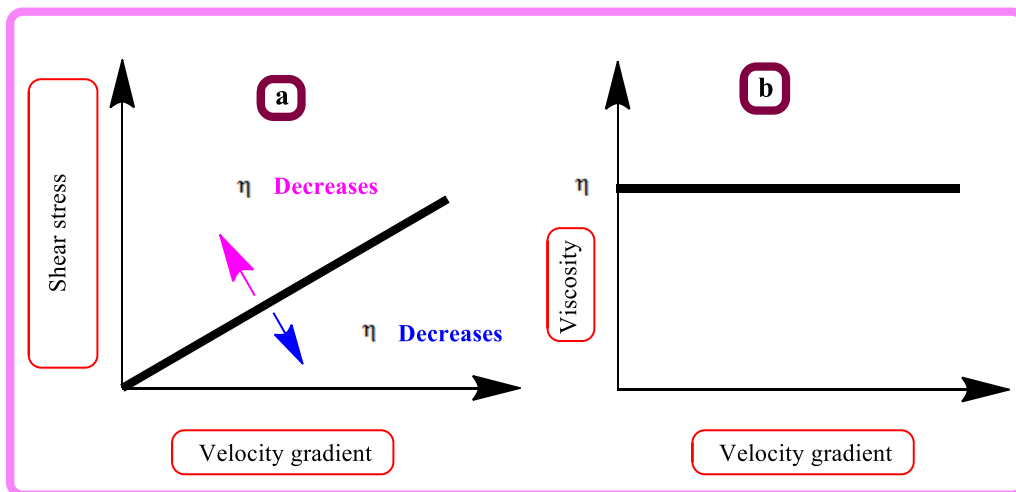


Figure 3. Viscosity of a Newtonian model.

2.3.2. Pseudo-plastic model.

Most polymers don't present the Newtonian models, this indicates that the viscosities are not independent of the shear rate however decrease according to the shear rate. The pseudo-plastic model of polymer signified that the viscosity decreases according to the shear rate. The rheological models proposed are a power law, law of tile, cross law, heat-dependent

pseudo-plastic, Williams-Landel-Ferry law, the law of Tile-Yasuda and Arrhenius law, respectively.

2.3.2.1. Power law.

The first behavior model for a non-carbonated fluid was developed by Ostwald-De Waele (Ostwald, 1923 and De Waele, 1923). It is written according to Equation 12 [54].

$$\eta = K \left| \dot{\gamma} \right|^{m-1} \quad (12)$$

Where K and m denote the consistency material and index pseudo-plasticity, respectively.

We observed that this law gives a good account of polymers' behavior with a high shear rate. It offers the advantage of authorizing analytical calculations in simple geometries. On the other hand, it has the disadvantage of not having a Newtonian plateau and even of leading to an infinite viscosity at zero shear rates [54].

2.3.2.2. Law of Tile.

The Power Law model applies over a limited range of the viscosity curve. In reality, at low shear rates, the polymers have a Newtonian behavior, i.e., the viscosity is independent of the shear rate. Carreau has developed a model that completely represents the viscosity curve and supports this plateau; its model is written according to Equation 13 [55].

$$\eta = \eta_0 \left[1 + (\lambda \dot{\gamma})^2 \right]^{(m-1)/2} \quad (13)$$

η_0 , λ , and m are the zero shear viscosity, the time constant, and the index of the law of power, respectively.

Hieber and Chiang (1989) proposed a complete review of its various laws and their applications to the case of the main polymers.

2.3.2.3. Cross Law.

The cross model is a variant of the square model; it is written according to Equation 14.

$$\eta - \eta_\infty = \frac{(\eta_0 - \eta_\infty)}{1 + (\lambda \dot{\gamma})^m} \quad (14)$$

With λ , η_0 and η_∞ are a characteristic time, the viscosity of the Newtonian plateau at the low shear rate and second Newtonian plateau at a very high shear rate, respectively. The change in viscosity according to the shear rate depends on the number of parameters to be adjusted.

2.3.2.4. Heat-dependent pseudo-plastic.

The heat-dependent pseudo-plastic considers the variation in viscosity according to the shear rate and temperature. An elevate in temperature suggesting a decrease in viscosity due to the polymer chains' increased mobility. In order to investigate the temperature effect on

viscosity, it is interesting to present the plot of viscosity according to shear stress for varying values of temperature. It is possible to drag all viscosities plots along a line at a constant shear rate to obtain a single plot. We can use the slip factor (a_T), which is according to the temperature; it has shown the difference between a viscosity plot at temperature (T) and the viscosity plot at the reference temperature (T_{ref}). a_T can be calculated from an Arrhenius law for semi-crystalline polymers or from the equation of Williams-Landel-Ferry (WLF) for polymers amorphous to a temperature between T_g and $T_g + 100$ °C (Equation 15) [54, 55].

$$a_T = \exp \left[\frac{E}{R} \left(\frac{1}{T} - \frac{1}{T_{ref}} \right) \right] \quad (15)$$

a_T , E , R , T , and T_{ref} are the slip factor, the activation energy, the molar constant of ideal gases, the temperature, and the reference temperature, respectively.

2.3.2.5. Williams-Landel-Ferry law (WLF).

The empirical Williams-Landel-Ferry relation (relation or WLF law), associated with the principle of time-temperature equivalence, makes it possible to account for variations in the limiting viscosity of amorphous (non-crystalline) polymers according to the temperature. The WLF law also expresses the variation with the temperature of the translation factor. Mathematical processing calculates a_T for each of the components M' and M'' of the measured complex modulus M^* . A good correlation between the two translation factors gives the values of the coefficients C_1 and C_2 characteristic of the material. The WLF law is only verified in the approximate temperature range $[T_g, T_g + 100$ °C] (Equation 16) [56].

$$a_T = \frac{\eta_0(T)}{\eta_0(T_{ref})} = \exp \left[\frac{-C_1(T - T_0)}{C_2(T - T_0)} \right] \quad (16)$$

The coefficients (or constants) C_1 and C_2 (positive) depend on the polymer considered and on T_0 as a reference temperature suitably chosen.

2.3.2.6. Law of Tile-Yasuda.

To obtain viscosity at other temperatures, we used the time-temperature superposition principle (Equation 17) [54, 55].

$$\dot{\eta}_{Tréf}(\gamma) = \dot{\eta}_0 \left[1 + (\lambda \dot{\gamma})^2 \right]^{(m-1)/2} \quad (17)$$

2.3.2.7. Arrhenius law.

The activation energy was introduced by the Swedish scientist Svante August Arrhenius in 1889. After having noted the empirical law that bears his name and describes the changes in viscosity with temperature. The activation energy was determined according to the following Equations 18 and 19.

$$\eta = \eta_{Tref} \exp \left[\frac{E}{R} \left(\frac{1}{T} - \frac{1}{T_{ref}} \right) \right] \quad (18)$$

$$\eta = \eta_0 \exp \left(\frac{E}{RT} \right) \quad (19)$$

T, E, R and η_0 are the temperature, the activation energy, the constant of ideal gases, and the constant.

3. Conclusions

This comprehensive review concerning the theoretical approach to rheological behavior allowed us to understand the phenomena relate to viscosimetric properties according to three models such as Newtonian model, pseudo-plastic model and pseudo-plastic thermo-dependent model, respectively.

The advantages of which are the determination of different thermodynamic parameters such as the activation energy (E_a), the variation of the activation standard enthalpy (ΔH_a), the variation of the activation standard entropy (ΔS_a), and the variation of the free activation standard energy (ΔG_a).

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Conflicts of Interest

The authors declare no conflict of interest.

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