

An In-silico Analysis of the Targeting of Cyclooxygenase-2, Lipoxygenases, and Glutathione S-transferase, Major Components of the Flesh of the Snail *Helix aspersa* Müller, Using a Molecular Docking: Research Focused on Colorectal Cancer

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Abstract: A cumulative accumulation of genetic and epigenetic abnormalities affecting the epithelial cells of the colon and/or the rectum causes colorectal cancer (CCR), a major cancer on a global scale. This study uses an in silico molecular docking approach to investigate the potential therapeutic benefits of bioactive extracts from the flesh of *Helix aspersa* Müller in the treatment of CCR. The toxicological analysis revealed that the chosen compounds had an acceptable safety profile with no adverse effects on hepatotoxicity, cardiotoxicity, or nephrotoxicity. The ADMET and biodisponibility investigations verified that all ligands met Lipinski's criteria, indicating their potential as drug candidates. Pharmacokinetic properties have shown that some compounds, such as octanoic acid and hexadecanoic acid, have high gastrointestinal absorption and the ability to pass through the blood-brain barrier, which may be advantageous for some applications. Evaluation of the liaison energies and inhibition constants between the ligands and the target proteins has been made possible by molecular docking. The findings demonstrated that for all target proteins, the most negative interaction energies and the lowest inhibition constants were provided by (3 β)-Cholest-5-en-3-ol and Cholest-4-en-3-one, indicating a strong affinity and a high potential inhibitor. Through hydrogen, alkyl, and pi-sigma bonds, these two compounds have also shown persistent interactions with key residues of the protein's active regions. Octanoic acid, on the other hand, has demonstrated the least affinity and the least amount of inhibitory power.

Keywords: CCR; *Helix aspersa* Müller; molecular docking.

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1. Introduction

Malignant diseases known as colorectal cancer (CCR) are characterized by the anarchic proliferation of glandular epithelial cells in the colon and/or the rectum. According to data from the World Health Organization, it is the third most diagnosed cancer and the second leading cause of death globally, making it a significant public health concern. Three main subtypes

have been identified: sporadic, inherited and linked to a chronic inflammatory condition [1]. According to Eng and Hochster [2], the incidence of this disease is still rising globally, with a noteworthy trend among young people in industrialized nations.

The established risk factors include certain lifestyle factors, such as imbalanced eating, sedentary lifestyles, excessive alcohol and tobacco use, genetic predispositions, and chronic pro-inflammatory conditions [3]. According to Ferlay et al. [4], 9.4% of cancer-related deaths in the environment were directly related to CCR.

The identification and management of modifiable risk factors play a critical role in primary prevention. Interventions based on improving lifestyle choices, such as increasing physical activity, adopting a healthy diet, and using preventive medications like aspirin, have been shown to be effective in lowering the incidence of CCR. In addition, systematic screening that enables the early detection and removal of precancerous lesions remains a crucial secondary strategy to slow the disease's progression [5-7].

There is ample evidence linking the CCR to chronic inflammatory bowel diseases like Crohn's disease and hemorrhagic rectocolitis. This link would be supported by accelerated cellular renewal and an increase in sporadic mutations brought on by persistent intestinal mucus inflammation [8].

The resistance to chemotherapeutic regimens seen in the majority of individuals with metastatic CCR represents another significant therapeutic challenge. Despite significant advancements in modern treatments that have improved global survival, pharmacological resistance continues to be a major barrier limiting the effectiveness of current strategies [9,10].

The current treatment of CCR is based on a multimodal approach that combines surgery, chemotherapy, targeted therapy, immunotherapy, and genetic therapy. The standard chemotherapeutic schemes, such as FOLFOX and CAPIRI, are frequently supplemented by specific targeted agents, such as Cetuximab or Bevacizumab, that interfere with key molecular pathways involved in tumor growth. Immunotherapy, including by inhibitors of immune control points (Pembrolizumab, Nivolumab), offers promising prospects for restoring the antitumoral immune response. Newly emerging approaches, such as adoptive T lymphocyte therapy and cytokine-based therapies, are actively being investigated to overcome the limitations of conventional treatments [11].

In this regard, natural products play a significant role in the search for new therapeutic options [12]. Approximately 50% of anticancer agents currently on the market are natural or derived from biodiversity-derived compounds [13,14]. These include various classes of biomolecules, such as alkaloids, polyphenols, polysaccharides, and diterpenoids, each of which has unique pharmacological characteristics.

The current study aims to investigate the potential of CCR inhibitors of natural bioactive compounds extracted from the flesh of *Helix aspersa* Müller. Using an in silico molecular docking approach, we assessed the interactions between three key proteins implicated in the physiopathology of CCR: glutathione S-transferase (GST), cyclooxygenase-1 (COX-1), and lipoxygenases (LOX), and seven bioactive compounds (L₁-L₇). This research may reveal new natural ligands that can block harmful pathogenic pathways, providing novel insights into the creation of alternative therapies for treating CCR.

2. Materials and Methods

2.1. Preparation of ligands.

According to our previous scientific research [15], seven bioactive compounds present in the snail flesh *Helix aspersa* Müller have been identified and chosen as ligands. The corresponding structures were extracted from the PubChem-NCBI database in 3D SDF format (Figure 1). PyMOL was used to convert the ligands from the SDF files to the Protein Data Bank (PDB) format, and the ligand molecules were downloaded separately in the AutoDock tools before being saved in the pdbqt format.

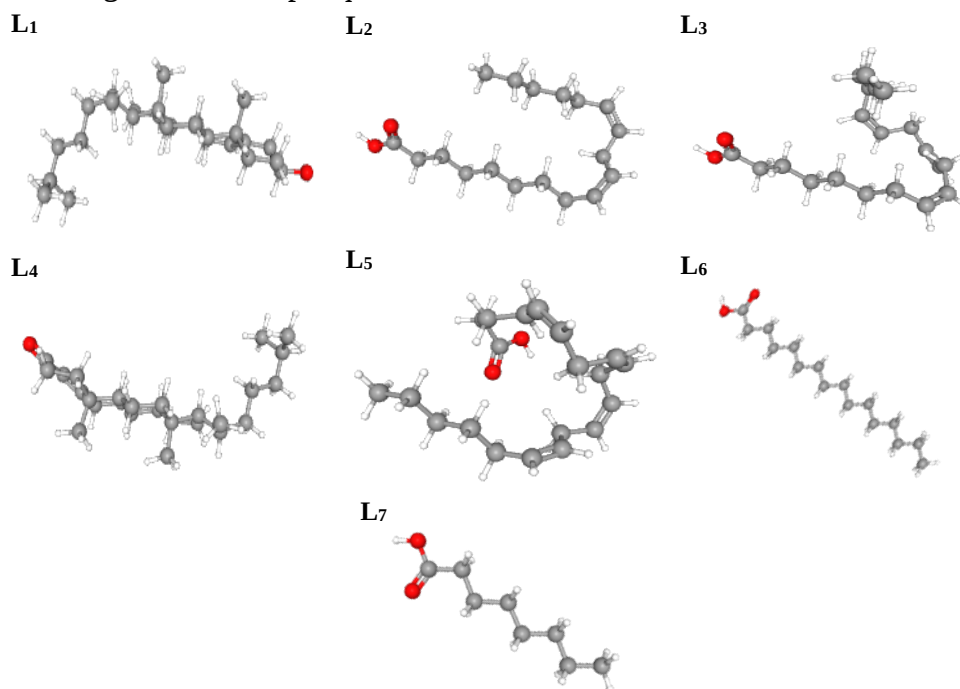


Figure 1. 3D structure of the ligands. (3 β)-Cholest-5-en-3-ol (**L1**); (Z,Z) 9,12-Octadecadienoic acid: (**L2**); 8,11,14-Eicosatrienoic acid: (**L3**); Cholest-4-en-3-one: (**L4**); cis-5,8,11,14-Eicosatetraenoic acid: (**L5**); Hexadecanoic acid (**L6**); Octanoic acid (**L7**).

2.2. Toxicity analysis.

The toxicity evaluation was conducted using Web ProTox-III [16], a program that is crucial for the development of new drugs because it predicts the toxicity of small molecules. The toxicity analysis carried out in this study took into consideration a number of factors, including the LD₅₀, the estimated class of toxicity, immunotoxicity, neurotoxicity, hepatotoxicity, mutagenicity, carcinogenicity, and cytotoxicity.

2.3. ADMET studies.

The ADMET analysis (adsorption, distribution, metabolism, excretion, and toxicity) is crucial to establish the pharmacodynamic characteristics of the molecule. SWISSADME, a web-based online server, was used to identify the therapeutic properties of natural compounds and drugs with the most likely matches [17]. In SWISSADME, the smiles of the ligands have been retrieved from PubChem.

2.4. Preparation of proteins.

The Protein Database (PDB) provided the crystalline structures of glutathione S-transferase (PDB ID: 7XBA), lipoxygenase-2 (PDB ID: 7LAF), and COX-1 (PDB ID: 6Y3C) (Figure 2). It was then incorporated into AutoDock, a molecular docking program. Using the Biovia Discovery Studio 2024 software (Accelrys, San Diego, CA, USA), the protein was first prepared by removing the ligand molecule from the active site. Next, the water molecules were removed in accordance with the indications found in the literature [18-20]. Then, hydrogen atoms were added to the enzyme structures using AutoDockTools 4.2.6 version 1.5.7. Partial atomic charges (Kollman charges) were then applied to the macromolecule, and it was turned into a PDBQT file.

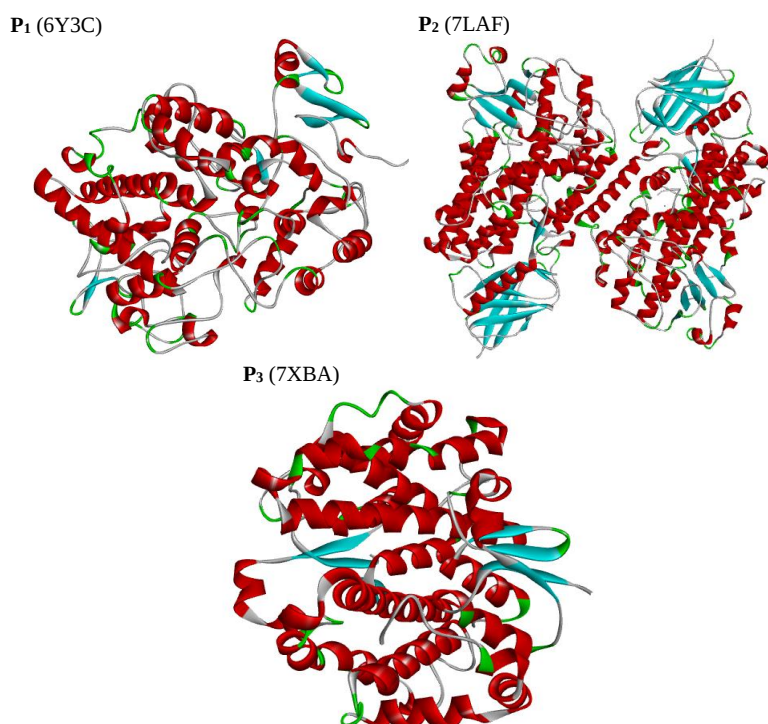


Figure 2. 3D structure of the proteins. **7XBA**: glutathione S-transferase, **7LAF**: lipoxygenase-2, and **6Y3C**: COX-1.

2.5. Molecular docking analysis.

In order to define the areas of interest in the enzyme, the AutoGrid software was used to perform docking calculations [21] using the genetic algorithm and the local search parameters (LGA). The standard script method was used to perform the molecular bond analysis of all the chosen bioactive compounds using AutoDock Vina [22]. The grid's dimensions were recorded in a file in the config.format.txt.

The following parameters were used to create a mesh centered on the protein's co-crystallized ligands: a mesh spacing of 0.500, a resolution of 60 Å 60 Å 60 Å points, and coordinates $x=-36.654$, $y=-51.733$, $z=2.080$; $x=-46.971$, $y=6.523$, $z=532.310$; $x=21.787$, $y=-22.229$, $z=-24.301$, respectively, for COX-1, LOX, and GST.

The Biovia discovery lab (Ver. 2024) has been used to study the fixed positions of complex structures and the aminated acids that interact.

3. Results and Discussion

3.1. Toxicity analysis.

Toxicity features were studied using ProTox-III. The prediction models are constructed from investigations carried out in vitro (Tox21 tests, bacterial mutation experiments of Ames, HepG2 cytotoxicity tests, and immunotoxicity tests) and in vivo (carcinogenicity, hepatotoxicity) (Table 1).

Table 1. Toxicity analysis of biocompounds.

L. No	LD ₅₀ / toxicity class	Neph	Hepat	Carcin	Card	Imm	Cyto	Neur
L ₁	890mg/kg; 4	0.92	0.85	0.51	0.81	0.99	0.95	0.51
L ₂	10000mg/kg; 6	0.91	0.55	0.55	0.99	0.96	0.71	0.91
L ₃	10000mg/kg; 6	0.55	0.55	0.64	0.99	0.96	0.71	0.91
L ₄	2300mg/kg; 5	0.91	0.77	0.57	0.83	0.99	0.81	0.56
L ₅	10000mg/kg; 6	0.55	0.55	0.64	0.99	0.98	0.71	0.91
L ₆	900mg/kg; 4	0.53	0.52	0.63	0.99	0.99	0.74	0.92
L ₇	900mg/kg; 4	0.53	0.52	0.63	0.99	0.99	0.74	0.92

Hepat: Hepatotoxicity; Carcin: Carcinogenicity; Neur: Neurotoxicity; Imm: Immunotoxicity; Cytot: Cytotoxicity; Neph: Nephrotoxicity; Card: Cardiotoxicity.

The bioactive compounds chosen for this study have not been found to be unreliable in terms of nephrotoxicity, hepatotoxicity, cardiotoxicity, or cytotoxicity. Similarly, L₁ has shown a dynamic activity in terms of carcinotoxicity, neurotoxicity, and immunotoxicity, and a noteworthy activity of L₄ has been observed in terms of neurotoxicity and immunotoxicity. The table shows the problems expected by the toxicity evaluation.

3.2. ADMET analysis.

All of the biocompounds met most of the parameters; therefore, the seven substrates met Lipinski's five criteria and were put through coupling studies (Table 2).

Table 2. Dependent variables of chemical compounds.

L. No	PubChem ID	LogP	MW	nOH	nOHNH	Nb	Nb viol	Mol ref
L ₁	5997	7.39	386.65	1	1	5	1	123.61
L ₂	5280450	5.88	280.45	2	1	14	1	89.46
L ₃	5280581	6.44	306.48	2	1	15	1	98.60
L ₄	91477	7.60	384.64	1	0	5	1	122.65
L ₅	444899	6.22	304.47	2	1	14	1	98.13
L ₆	985	5.55	256.42	2	1	14	1	80.80
L ₇	379	2.43	144.21	2	1	6	0	42.34

LogP: lipophilicity; Mol Wt: molecular weight; nOH: no. of H bond acceptors; nOHNH: no. of H bond donors; Nb: no. of rotatable bonds; Nb viol: no of violations; Mol ref: molecular refractivity.

A compound's hydrophobicity is represented by logP; a high lipophilicity shown by a considerable logP (logP >5) will probably improve the permeability of the membranes, but it can also result in lowered water solubility. High values for L₁ (7.39), L₃ (6.44), and L₄ (7.60) point to a high lipophilicity that would enhance the bridging of membranes while perhaps affecting solubility. Still, L₇ (2.43) has the lowest logP, indicating more hydrophilicity, which would probably enhance solubility and lower permeability.

According to Lipinski's rule of five, compounds having a molecular weight less than 500 Da are usually regarded as therapeutic agents. The seven ligands satisfy this requirement; hence, they are possible candidates for the synthesis of medications.

Solubility and bond interactions are influenced by the number of hydrogen bond acceptors (nOH) and donors (nOHNH). L₄ has no hydrogen donors, which could decrease its ability to dissolve and interact. On the other hand, L₂, L₆, and L₇ have two hydrogen atoms that are bond donors, suggesting an ability to solubilize and interact with biological targets.

Molecular flexibility is influenced by the number of rotating bonds (Nb), which affects the efficiency of the bond. L₃ offers the greatest flexibility (15 rotating links), which gives it the potential ability to adjust to various connection points. L₇ has the lowest score (6), which indicates that it is more rigid.

Molecular refractivity (Mol ref), which denotes the polarizability of a molecule, influences its solubility and interactions. L₁ and L₄ display the greatest refractivity (123.61), suggesting intense interactions. L₇ has the lowest score (42.34), indicating a lower interaction potential.

3.3. Pharmacokinetic analysis and ligand comparisons.

The *in silico* pharmacokinetic profile of the ligands, as illustrated in Table 3, emphasizes that the drug similarity and the pharmacokinetic characteristics of certain compounds resemble those of the drugs already marketed.

Table 3. *In silico* pharmacokinetics of ligands using SwissADME.

L. No	ESOL (Log S)	Glads	BBB	P-gp	Cytochromes				
					3A4	1A2	2C19	2C9	2D6
L ₁	-7.40 (PS)	Low	No	No	No	No	No	Yes	No
L ₂	-7.40 (PS)	Low	No	No	No	No	No	Yes	No
L ₃	-7.40 (PS)	Low	No	No	No	No	No	Yes	No
L ₄	-7.17 (PS)	Low	No	No	No	No	No	Yes	No
L ₅	-7.17 (PS)	Low	No	No	No	No	No	Yes	No
L ₆	-5.02 (PS)	High	Yes	No	No	Yes	No	Yes	No
L ₇	-2.26 (S)	High	Yes	No	No	No	No	No	No

Table 3 lists the results of the ADME prediction for the seven compounds, including the cutaneous permeability value, the gastrointestinal absorption, the permeability of the BBB, the P-glycoprotein substrate, the inhibition capacity, and the substrate of the cytochrome P450 enzyme derived from the SwissADME prediction.

The ESOL Log S values represent the ability of the ligands to dissolve. Lower (more negative) values indicate a lower solubility. L₁ to L₆ has a very low solubility (-7.40 to -7.17), indicating their low dissolution capacity (PS). This could influence their bioavailability. L₇ (-2.26) has a higher solubility, being classified as soluble (S).

It is likely that the ligands having a high gastrointestinal absorption are effectively assimilated at the intestinal level. L₆ and L₇ have a high absorption capacity, which makes them more promising for oral bioavailability. The others (L₁-L₅) have low absorption, which could limit their effectiveness as oral treatments.

L₆ and L₇ have the ability to cross the Blood-brain Barrier, which indicates that they could exert an action at the level of the central nervous system. L₁-L₅ do not cross the blood-brain barrier, which limits their potential use in central nervous system applications.

The P-gp is a transporter that expels drugs out of cells, influencing the preservation of drugs. None of the ligands act as a substrate for P-gp, which indicates that they are not likely to be expelled, which is conducive to bioavailability.

CYP enzymes, present in the liver, intervene in the metabolism of drugs by influencing their degradation and elimination. CYP3A4: only L₆ is interacting, indicating that it could be

metabolized quickly. CYP1A2: L6 indicates an interaction, suggesting possible impacts on the metabolism of drugs. CYP2C19: no ligands show interaction with this enzyme, which indicates a low probability of its metabolism. CYP2C9: an interaction between L₁ and L₅ suggests a potential metabolism by this enzyme. CYP2D6: only the L₁ is affected, which could have an impact on its pharmacokinetics.

The interaction of cytochrome P450 isoenzymes with drugs can lead either to an accelerated metabolism when the drug functions as a substrate for a CYP, causing induction, or to an accumulation of the drug when the latter plays the role of inhibitor, thus resulting in inhibition. These two scenarios are generally perceived as having negative consequences [23]. Thus, in the context of drug development, *in silico* methods to anticipate the interactions between substances or drugs and CYP isoenzymes are of paramount importance.

3.4. Bioavailability radar.

The bioavailability tool offers a quick assessment of the similarity of a compound with an already established pharmaceutical agent. As illustrated in the graph, the magenta area defines the ideal range for each parameter. When analyzing the characteristics of a compound, its radar representation must be located in this magenta zone to be recognized as a drug. Therefore, the ligands must demonstrate oral bioavailability or lack thereof based on their radar representation.

The compound L₇ satisfies the criteria of the radar graph and can therefore, be suggested as being orally bioavailable (Figure 3).

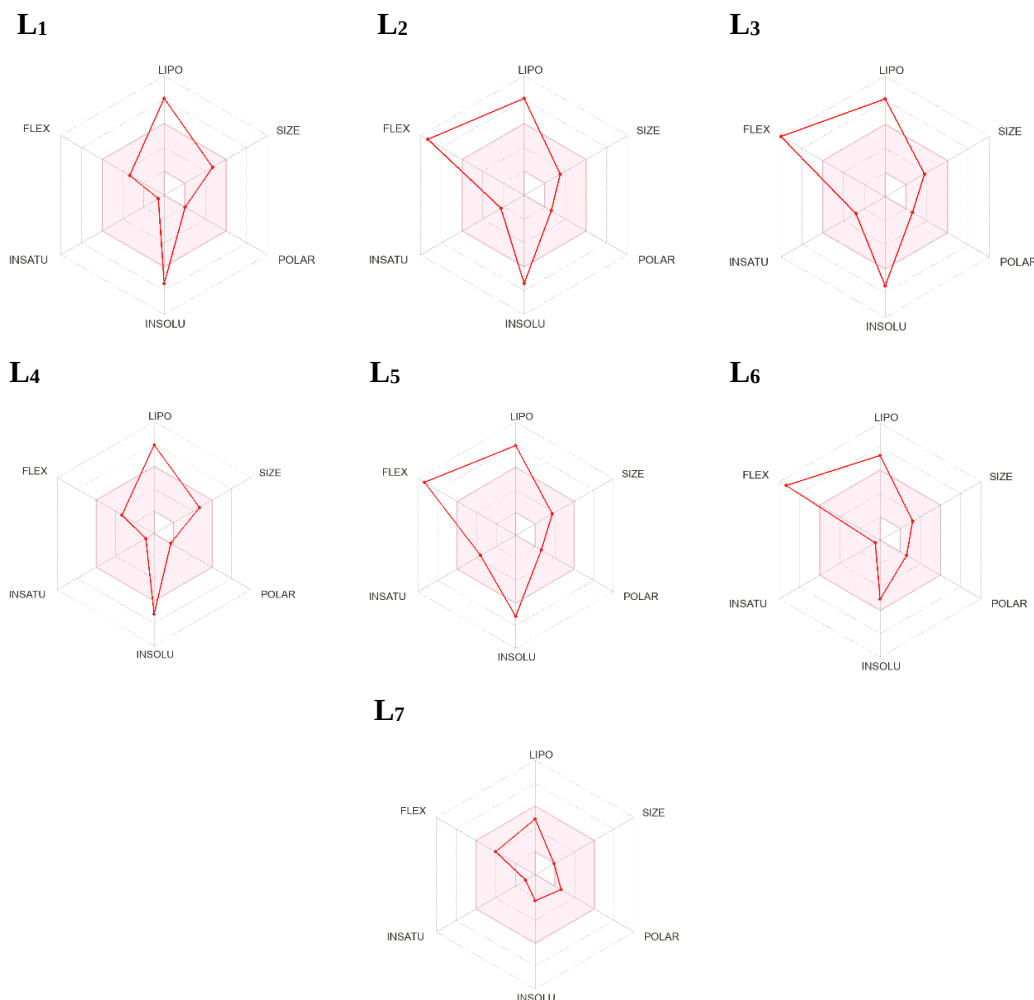


Figure 3. Radar graphs of the ligands.

3.5. Docking analysis.

A molecular docking analysis was carried out on all the ligands filtered from the ADME analysis. The molecular docking computer tool plays a crucial role in drug search. This is done in order to more accurately select possible compounds and examine the creation of bonds at the binding site in the protein-ligand complex. The binding energy influences the binding affinity between ligands and receptors; the lower the energy, the higher the binding affinity [24].

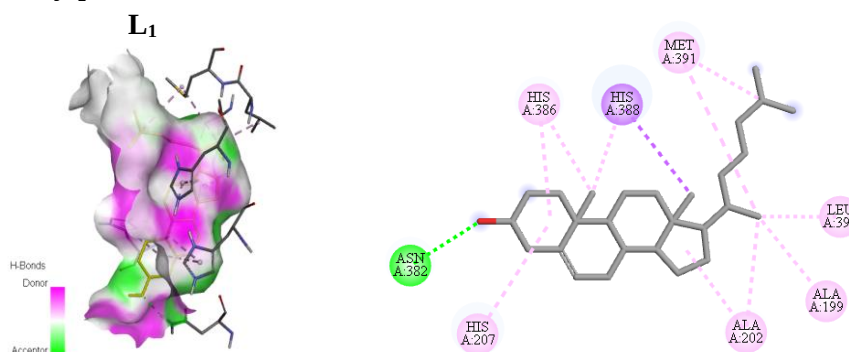
To achieve this goal, seven bioactive compounds were associated with proteins, and their binding energies were compared. The data relating to the binding energy (kcal/mol), to the inhibition constant, and to the amino acids involved in the bonds were recorded following the docking process (Tables 4, 5, and 6).

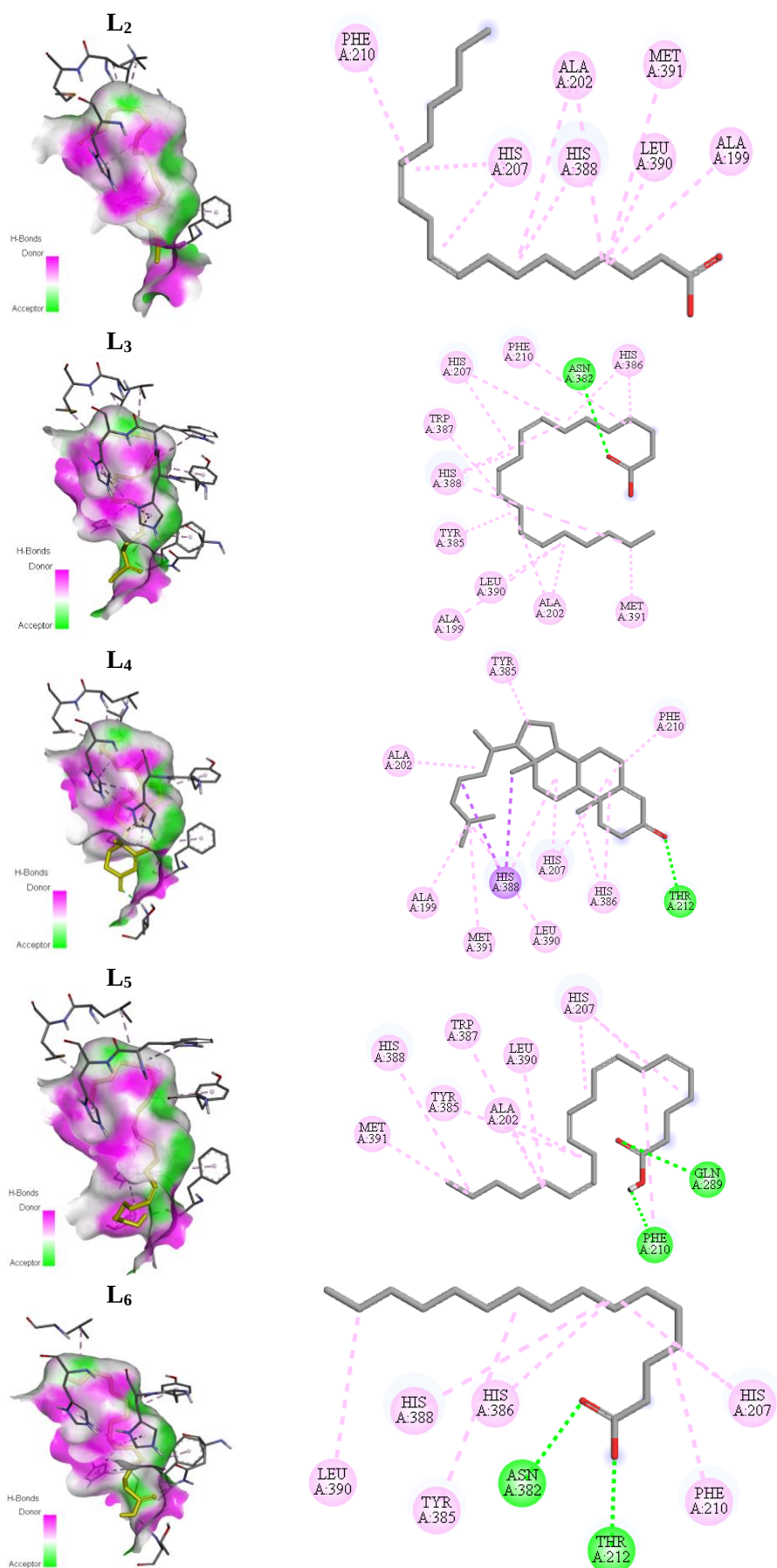
Table 4. Interaction scores between bioactive compounds and 6Y3C.

L. No	Be (Kcal/mol)	Ic (μ M)	Amino acids involved
L ₁	-11.45	4.06 nM	Asn A:382; Ala A:202; Ala A:199; Leu A :199; Leu A:390; His A:388; His A:386; Met A:391; His A:207
L ₂	-5.28	135.86	Phe A:210; His A:207; Ala A:202; His A:388; Met A: 391; Leu A:390; Ala A:199
L ₃	-6.61	14.20	Asn A:382; His A:386; Met A:391; Ala A: 210; Leu A:390; Phe A:210; His A:207; Trp A:387; His A:388; Tyr A:385; Ala A:199
L ₄	-11.27	5.52 nM	Ala A:202; Ala A:199; Tyr A:385; Met A:391; His A:388; Leu A:390; His A:207; Phe A:210; Thr A:212; His A:386
L ₅	-6.08	34.83	Gln A:289; Phe A:210; Met A:391; His A:388; Tyr A:385; Trp A:387; Ala A:202; Leu A:390; His A:207
L ₆	-5.16	165.40	Asn A:382; Thr A:212; His A:207; Phe A:210; Leu A:390; His A:388; His A:386; His A:385
L ₇	-4.63	400.67	Arg A:120; Trp A:387; Met A:522; Phe A:518; Ile A:523; Val A:349; Ala A:527

Be: Binding energy; Ic: Inhibition constant.

The ligands L₁ and L₄ display the most negative binding energies, respectively, -11.45 and -11.27 Kcal/mol, as well as the lowest Ic values, namely 4.06 nM and 5.52 nM. This means they have the greatest affinity towards the target protein and are the most powerful inhibitors. The ligands L₂, L₃, L₅, L₆, and L₇ have less negative binding energies as well as higher Ic, indicating lower affinity and inhibition. The L₇ ligand has the lowest affinity and the most modest inhibitory power.





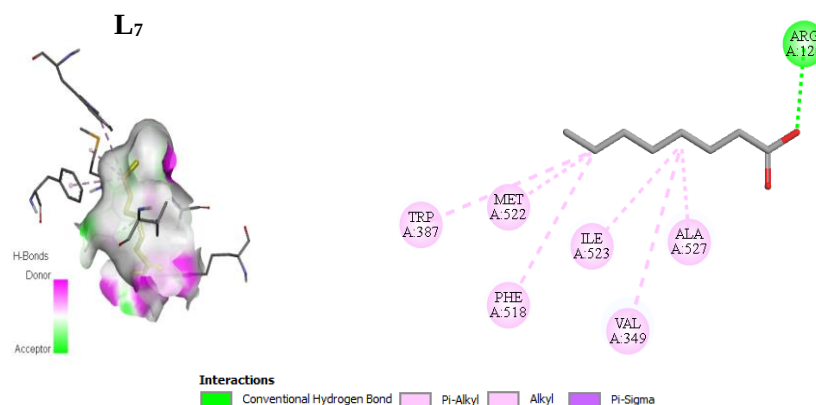


Figure 4. Binding mode of L₁-L₇ compounds with the 6Y3C receptor 3D (left) and 2D (right) views.

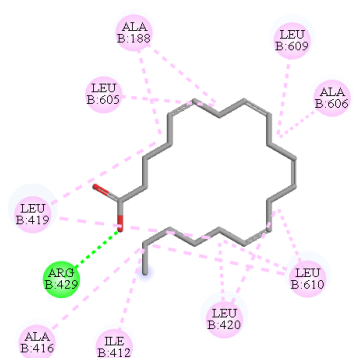
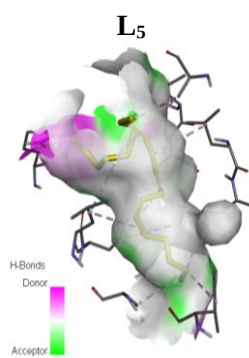
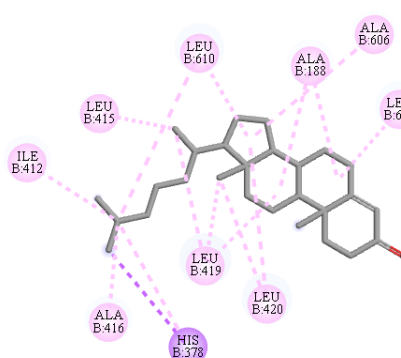
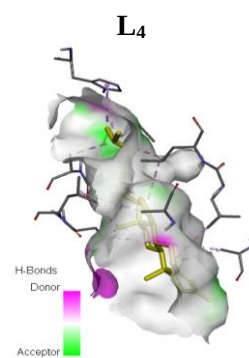
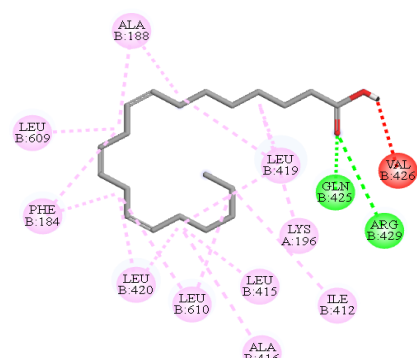
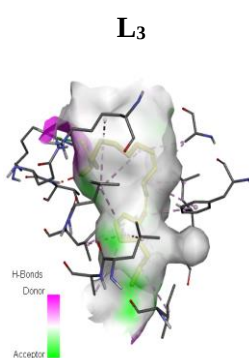
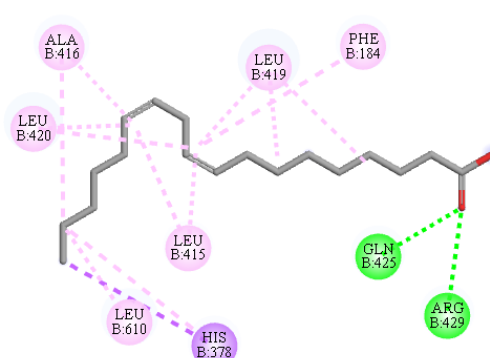
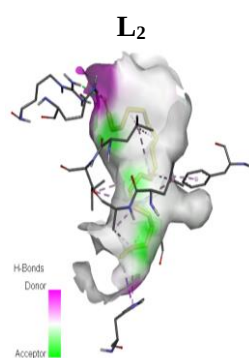
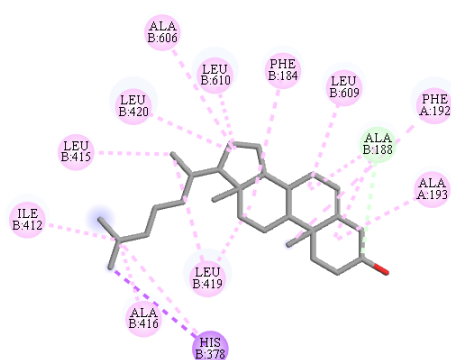
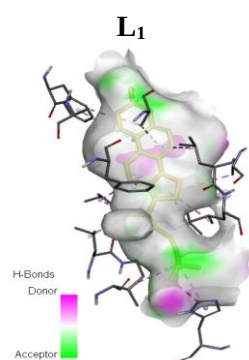
The amino acids involved indicate the types of interactions that take place. The results of the molecular docking study showed that the compound L₁ had a hydrogen bond interaction with the protein residues Asn A:382 and an alkyl-type interaction with the protein residues His A:207, His A:386, Leu A:390, Met A:391, Ala A:199 and Ala A:202, and a pi-sigma bond with the protein residues His A:388. In addition, the compound L₄ showed an interaction at the level of a hydrogen bond with the protein residue Thr A:212, as well as eight alkyl bonds with the protein residues Tyr A:385, Phe A:210, Ala A:202, His A:207, His A:386, Leu A:390, Met A:391 and Ala A:199 and a pi-sigma bond with the protein residue His A:388. However, the compound L₇ showed an interaction at one hydrogen bond with the protein residues Arg A:120, and six alkyl bonds with the protein residues Trp A:387, Met A:522, Phe A:518, Ile A:523, Val A:349, and Ala A:527 (Table 4, and Figure 4).

Table 5. Interaction scores between bioactive compounds and 7LAF

L. No	Be (Kcal/mol)	Ic (μM)	Amino acids involved
L ₁	-10.42	23.17 nM	Ala B:606; Leu B:610; Phe B:184; Leu B:609; Leu B:420; Phe A:192; Leu B:415; Ala B:188; Ala A:193; Ile B:412; Leu B:419; Ala B:416; His B:378
L ₂	-5.53	88.40	Ala B:416; Leu B:420; Leu B:419; Phe B:184; Leu B:415; Leu B:610; Gln B:425; His B:378; Arg B:429
L ₃	-6.16	30.45	Ala B:188; Leu B:609; Leu B:419; Val B:426; Gln B:425; Phe B:184; Lys A:196; Arg B:429; Leu B:420; Leu B:610; Leu B:415; Ile B:412; Ala B:416
L ₄	-11.05	7.99 nM	Leu B:610; Ala B:606; Ala B:188; Leu B:415; Leu B:609; Ile B:412; Leu B:419; Ala B:416; Leu B:420; His B:378
L ₅	-5.58	81.12	Ala B:188; Leu B:609; Leu B:605; Ala B:606; Leu B:419; Arg B:429; Leu B:610; Ala B:416; Ile B:412; Leu B:420
L ₆	-4.89	259.93	Phe B:184; Leu B:610; Ile A:197; Ala B:188; Lys A:196; Leu B:606; Leu B:419; Ala B:609; Phe A:192; Arg B:429; Val B:426; Leu B:420
L ₇	-4.63	400.67	Val B:426; Leu B:419; Leu B:420; Gln B:425; Arg B:429; Leu B:610

Be: Binding energy; Ic: Inhibition constant.

The ligands L₁ and L₄ have the most negative binding energies (-10.42 and -11.05 Kcal/mol, respectively) and the lowest Ic (23.17 nM and 7.99 nM). This indicates they have the highest affinity for the target protein and are the strongest inhibitors. The ligands L₂, L₃, L₅, L₆, and L₇ have less negative binding energies and higher Ic, which suggests lower affinity and inhibition. The L₇ ligand is the one with the least affinity and the lowest inhibitory power (Table 5).



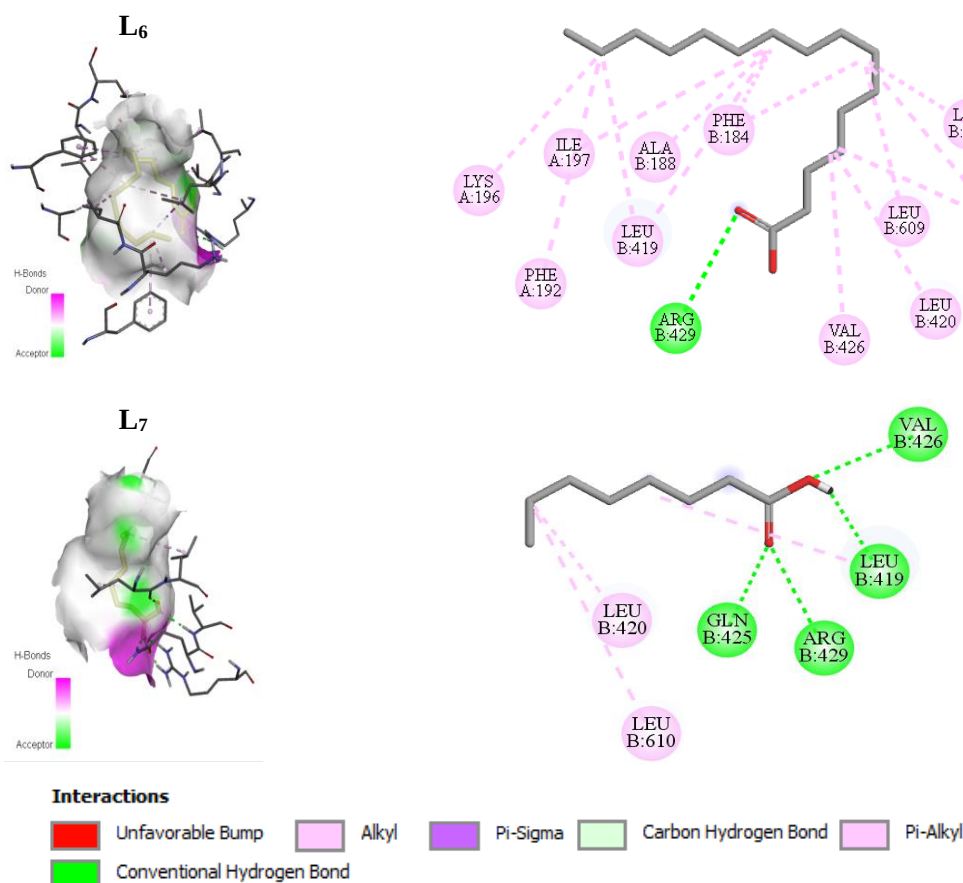


Figure 5. Binding mode of L₁-L₇ compounds with the 7LAF receptor 3D (left) and 2D (right) views.

The results of the molecular docking study showed that the compound L₁ had a pi-sigma bond interaction with the protein residue His B:378, a hydrogen carbon bond with the protein residue Ala B:188, and eleven alkyl-type interactions with the protein residues Ala B:606, Leu B:610, Phe B:184, Leu B:609, Leu B:420, Phe A:192, Leu B:415, Ala A:193, Ile B:412, Leu B:419 and Ala B:416. In addition, the compound L₄ showed an interaction at the level of a carbon-hydrogen bond with the protein residue His B:378, as well as nine alkyl bonds with the protein residues Leu B:610, Ala B:606, Ala B:188, Leu B:415, Leu B:609, Ile B:412, Leu B:419, Ala B:416 and Leu B:420. However, the compound L₇ showed four interactions at a hydrogen bond with the protein residues Val B:426, Leu B:419, Gln B:425, and Arg B:429, and two alkyl bonds with the protein residues Leu B:420 and Leu B:610 (Figure 5).

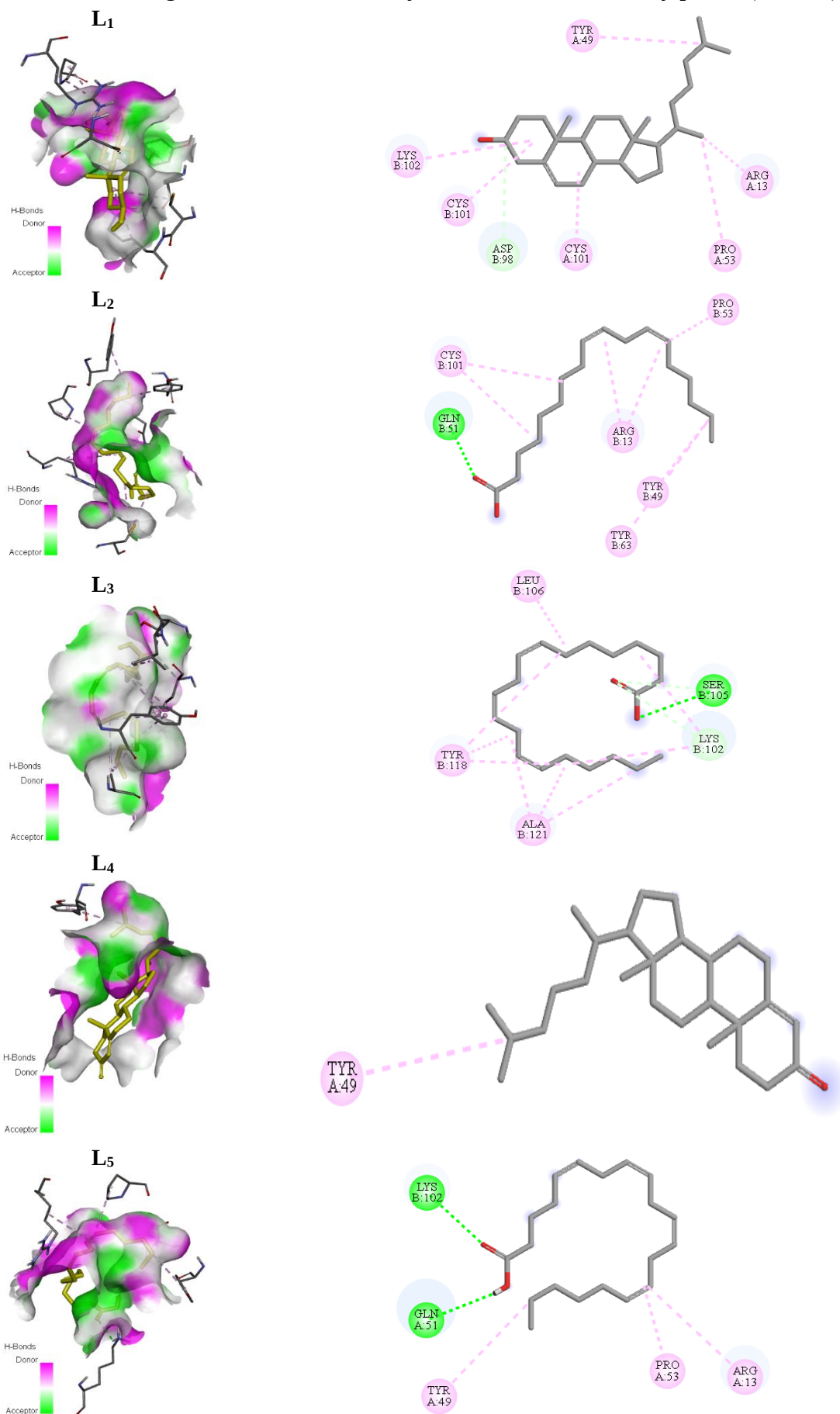
Table 6. Interaction scores between bioactive compounds and 7XBA.

L. No	Be (Kcal/mol)	Ic (μM)	Amino acids involved
L ₁	-7.12	6.00	Tyr A:49; Lys B:102; Arg A:13; Cys B:101; Asp B:98; Cys A:101; Pro A:53
L ₂	-3.40	3.23 mM	Pro B:53; Cys B:101; Gln B:51; Arg B:13; Tyr B:49; Tyr B:63
L ₃	-3.10	5.32 mM	Leu B:106; Ser B:105; Tyr B:118; Lys B:102; Ala B:121
L ₄	-7.30	4.45	Tyr B:49
L ₅	-3.97	1.24 mM	Lys B:102; Gln A:51; Pro A:53; Arg A:13; Tyr A:49
L ₆	-3.80	1.63 mM	Tyr A:49; Lys B:102
L ₇	-4.53	475.21	Arg B:13; Pro B:53

Be: Binding energy; Ic: Inhibition constant

The ligands L₁ and L₄ have the most negative binding energies (-7.12 and -7.30 Kcal/mol, respectively) and the lowest Ic (6.00 μM and 4.45 μM). This indicates they have the highest affinity for the target protein and are the strongest inhibitors. The other ligands have

less negative binding energies and higher I_c (in mM), which suggests lower affinity and inhibition. The L₃ ligand has the least affinity and the lowest inhibitory power (Table 6).



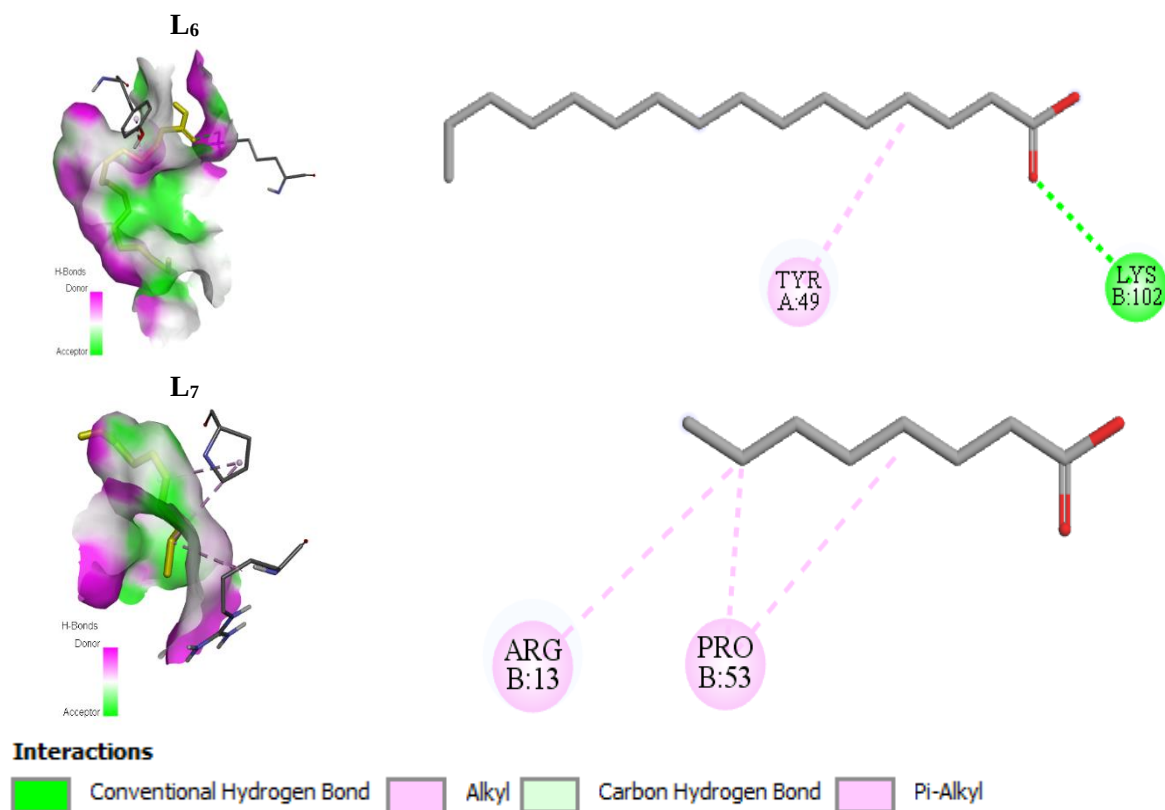


Figure 6. Binding mode of L₁-L₇ compounds with the 7XBA receptor 3D (left) and 2D (right) views

The results of the molecular docking study showed that the compound L₁ had a hydrogen carbon bond with the protein residue Asp B:98 and six alkyl-type interactions with the protein residues Tyr A:49, Lys B:102, Arg A:13, Cys B:101, Cys A:101, and Pro A:53. In addition, the compound L₄ presented an alkyl bond with the protein residues Tyr A:49. However, the compound L₃ showed an interaction at a hydrogen bond with the protein residue Ser B:105, a carbon-hydrogen bond with the protein residue Lys B:102 and three alkyl bonds with the protein residues Leu B:106, Tyr B:118, and Ala B:121 (Figure 6).

The L₁ and L₄ ligands could be promising candidates for developing drugs targeting this protein. A more in-depth study of their interactions and biological effects would be necessary.

4. Conclusions

This study highlights the potential of natural bioactive extracts of *Helix aspersa* Müller as potential inhibitors in treating colorectal cancer. The *in silico* results of this study indicate that the bioactive extracts of the flesh of *Helix aspersa* Müller, specifically (3 β)-cholest-5-en-3-ol and Cholest-4-en-3-one, have promising therapeutic potential for the treatment of colorectal cancer. Their strong affinity for the protein targets COX-1, LOX, and GST, along with their safety profile and advantageous pharmacokinetic properties, make them intriguing candidates for creating novel therapeutic approaches. Although these results are encouraging, more *in vitro* and *in vivo* experimental research is required to confirm these compounds' biological effectiveness, selectivity, and safety. Furthermore, more research on their mechanisms of action, biodisponibility, and metabolic stability would help to understand their potential in clinical settings better. Thus, this research opens up new avenues for the use of natural products in the creation of alternative treatments for colorectal cancer.

Author Contributions

Conceptualization, M.A., and M.Z.; methodology, M.A.; software, M.A., and M.Z.; validation, M.A., A.R., and R.B.; formal analysis, M.A.; data curation, M.A., and M.Z.; writing—original draft preparation, R.B.; writing—review and editing, A.R., R.B.; supervision, R.B.; All authors have read and agreed to the published version of the manuscript.

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Data Availability Statement

Data supporting the findings of this study are available upon reasonable request from the corresponding author.

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Conflicts of Interest

The authors declare no conflict of interest.

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